### Microtubule-Stabilizing Marine Metabolite Laulimalide and Its Derivatives: Synthetic Approaches and Antitumor Activity

Johann Mulzer<sup>\*,†</sup> and Elisabeth Öhler\*

Institut für Organische Chemie der Universität Wien, Währinger Strasse 38, A-1090 Wien, Austria

Received March 5, 2003

#### Contents

. . . . ..

Ι.	Introduction	3/53
II.	Chronological Evolution of Laulimalide Synthesis	3754
III.	Early Synthetic Work	3755
	A. First Laulimalide Fragment (Ghosh)	3755
	B. Fragment Syntheses by Nishiyama	3755
	C. First RCM-Based Fragment Syntheses (Mulzer and Ghosh)	3757
IV.	Total Syntheses of Laulimalide and Analogues	3759
	A. First Total Synthesis of Laulimalide (Ghosh and Wang)	3759
	B. Stereocontrolled Introduction of the 2,3-Z-Enoate (Ghosh)	3761
	C. Last Step Introduction of the Epoxide	3761
	1. Mulzer's First Approach	3761
	<ol> <li>Synthesis of Laulimalide Analogues (Mulzer)</li> </ol>	3765
	<ol> <li>Ring Closure by Mitsunobu-Lactonization (Paterson's Approach)</li> </ol>	3765
	D. Chiral Allylsilane Addition	3767
	1. For Macrocyclization (Mulzer and Enev)	3767
	<ol><li>For Fragment Union (Mulzer and Hanbauer)</li></ol>	3769
	E. Regioselective Macrolactonization with Unprotected Diol (Wender)	3770
	F. Early Incorporation of the Epoxide	3772
	1. Crimmins' Approach	3772
	2. Williams' Approach	3774
	G. Asymmetric Acyl Halide–Aldehyde Cyclocondensation (Nelson)	3775
V.	Syntheses of Laulimalide Fragments	3777
	A. Davidson's Contributions	3777
	B. Mulzer's (S)-Citronellal Based Route to a $C_3-C_{14}$ Fragment	3779
	C. Syntheses of Laulimalide Subunits by Lee et al.	3780
VI.	General Evaluation	3781
	A. Construction of DHP Fragments	3781
	B. Connection of Main Fragments	3782
	C. Macrocyclization	3782
	D. The Endgame	3782
VII.	Antitumor Activity of Laulimalides and Analogues	3782
VIII.	Conclusion and Prospects	3783

<sup>†</sup> Phone: +431-4277-52190. Fax: +431-4277-52189. E-mail johann.mulzer@univie.ac.at.

IX.	Acknowledgments	3783
Х.	References and Notes	3784

#### I. Introduction

In 1988, Cacospongia mycofijiensis, a chocolate sponge collected from Vanuatu, became the object of an intense study at the University of California because the liquid squeezed from freshly collected material killed tropical fish being held in an aquarium within 10 min. This primary effort to protect the sponge against predators originated from two cytotoxic macrolides-fijianolide B (1) and fijianolide A (2)—whose gross structure was elucidated mostly by NMR analysis.<sup>1b</sup> Independently, the same compounds were isolated by Hawaiian scientists from an Indonesian sponge-Hyattella sp.-and given the now commonly used names laulimalide (1) and isolaulimalide (2),<sup>1a</sup> the names being derived from Hawaiian language ("laulima" = people working together), due to the cooperating research groups. Interestingly, 1 and 2 were also isolated from the extracts of a predator nudibranch, Chromodoriz lochi, that was found grazing on the sponge.<sup>1a,b</sup> Later on,  $\mathbf{1}$  and  $\mathbf{2}$ were also isolated from the Okinawan sponge *Fas-ciospongia rimosa*<sup>1d</sup> and very recently also from a sponge in the genus *Dactylospongia*.<sup>1e</sup> In 1996, the structure of **1** and its absolute configuration was confirmed by Higa et al. through X-ray diffraction studies.<sup>1c</sup> Higa's group also isolated a ring-expanded regioisomer of 1-neolaulimalide (3)-as a minor congener of 1 and 2.1d

Laulimalide (1), meanwhile identified as a potent inhibitor of cellular proliferation with  $IC_{50}$  values against numerous drug-sensitive cell lines in the low nanomolar range, <sup>1a,d,2</sup> is an 18-membered macrolide (within the inner perimeter) that contains nine chiral carbons (5*R*, 9*S*, 11*S*, 16*S*, 17*S*, 19*S*, 20*S*, 23*S*) and two dihydropyran rings, one ( $C_5-C_9$ ) annulated to the macrocycle in 2,6-trans fashion and the second one ( $C_{23}-C_{27}$ ) connected to the macrolide core via an *E*-allylic alcohol. Furthermore, **1** contains a transdisubstituted epoxide at  $C_{16}$  and  $C_{17}$  and a 2,3-*Z*enoate.

Isolaulimalide (2) is an isomer of 1 whose tetrahydrofuran ring is formed by an  $S_N2$ -type attack of the  $C_{20}$  hydroxyl group on  $C_{17}$  of the epoxide, the acidcatalyzed isomerization being complete within a few hours.<sup>1b</sup> Isolaulimalide exhibits significantly reduced



Johann Mulzer was born in 1944 at Prien in Upper Bavaria, Germany. In 1974 he received his Ph.D. degree under the supervision of Rolf Huisgen at the Ludwig-Maximilians University in Munich. Subsequently, he joined the group of E. J. Corey at Harvard as a postdoctoral fellow. From 1982 to 1996 he held professorships at the University of Düsseldorf, the Free University of Berlin, and the Johann-Wolfgang-Goethe University in Frankfurt. Since 1996 he has been a full professor at the Institute of Organic Chemistry of the University of Vienna. His main research interests are focused on the total synthesis of structurally and physiologically interesting natural products.



Elisabeth Öhler was born in Vienna, Austria, and received her Ph.D. degree in Organic Chemistry at the University of Vienna in 1968 for work on reactions with organophosphorus compounds under the guidance of Friedrich Wessely. Afterward, she joined Ulrich Schmidt for work on Reformatsky-type reactions and syntheses of dehydroamino peptides, and in 1978 she joined the team of Erich Zbiral to explore reactions of unsaturated phosphonates. Since 1996, she has been a senior research fellow in the Mulzer group. Apart from laulimalide, she has also contributed to various epothilone syntheses.

activity with  $IC_{50}$  values in the low micromolar range, which may indicate that the epoxide function is necessary for high activity.

Neolaulimalide (3) is the ring-enlarged regioisomer of 1 with an intact epoxide moiety and was reported to possess high cytotoxicity in the same range as 1.<sup>1d</sup> It is distinctly more stable than 1, the acid-promoted rearrangement to 2 being complete only after 2 days.<sup>1d</sup>

In February 1999, **1** and **2** were recognized as new members of the MSAA (microtubule-stabilizing antitumor agents) family of compounds,<sup>2a</sup> which share the same or a similar mechanism of action as the frontline anticancer drugs Taxol (paclitaxel)<sup>3</sup> and Taxotere (docetaxel). Thus, the list of compounds with "taxol-like" activity<sup>4</sup> currently includes the following members: taxanes (isolated from yew trees),

marine metabolites (sarcodictyins/eleutherobin, discodermolide, laulimalide, dictyostatin, and peloruside A),<sup>5</sup> microbial metabolites (the epothilones,<sup>6</sup> which are already under clinical investigation and the polycyclic compound FR182877, formerly known as WS9885B<sup>7</sup>), other natural products (taccalonolide,<sup>8a</sup> tryprostatin,<sup>8b</sup> xanthochymol<sup>8c</sup>), and non-natural compounds (for instance, an analogue of estradiol,<sup>9a</sup> a combretastatin D analogue<sup>9b</sup> and GS-164<sup>9c</sup>). Moreover, it was shown that **1**, like the epothilones and discodermolide, is an effective inhibitor of cell growth in paclitaxel-resistant cells.<sup>2</sup> A very recent study revealed that laulimalide, in contrast to the epothilones, discodermolide, and eleutherobin, apparently does not bind at the taxol site to the tubulin polymer and is also active against epothiloneresistant cell lines.<sup>2b</sup>

Apart from the significant clinical potential of **1** and its restricted natural supply, the attraction of laulimalide as a synthetic target originates from its unique and complex molecular architecture. Specifically, its 16,17-epoxide is susceptible to nucleophilic attack from the 20-hydroxy group to form the more stable tetrahydrofuran isomer **2** and the 2,3-*cis*-enoate moiety readily undergoes Z/E-isomerization. In the following, it will be demonstrated that the principal difficulties during total synthesis of laulimalide arise during or after the introduction of these two functionalities.

To date, the use of ring-closing olefin metathesis (RCM) in laulimalide synthesis<sup>10a</sup> and selected total syntheses of  $1^{10b}$  have been briefly reviewed. The present review will summarize the complete laulimalide-directed synthetic work available up to April 2003, including the syntheses of some non-natural analogues, as well as the biological data available from laulimalide and its natural co-metabolites **2** and **3** and from the few analogues which have been investigated to date.

#### *II. Chronological Evolution of Laulimalide Synthesis*

Synthetic work toward laulimalide<sup>11–23</sup> started in 1996, when the absolute configuration of 1 was determined.<sup>1c</sup> These early efforts resulted in three reports on fragment syntheses by the groups of Ghosh<sup>11a</sup> and Nishiyama.<sup>12</sup> However, these primary approaches were not successfully brought to completion. The interest in 1 was distinctly intensified after February 1999, when laulimalide was identified as a new member of the MSAA family.<sup>2a</sup> In September 1999, Mulzer's group reported the synthesis of the "lower"  $C_1 - C_{12}$  moiety of **1**, utilizing ring-closing olefin metathesis (RCM)<sup>24</sup> for the construction of the dihydropyran subunit.13a This communication was followed in close succession by an independent report of Ghosh and Wang, concerning the synthesis of an extended  $C_2-C_{16}$  fragment of 1 which used a slightly different RCM methodology for elaboration of the dihydropyran moiety.<sup>11b</sup>

Since these early fragment syntheses, an impressive number of 16 approaches to key fragments of **1** have been contributed by different groups,<sup>11–16</sup> and seven teams have completed as many as 10 total



**Figure 1.** Laulimalide (1), isolaulimalide (2), and neolaulimalide (3).



**Figure 2.** Unsuccessful or nonstereoselective ring closures, and selective epoxidation of unprotected desepoxylaulimalide.

syntheses.<sup>17–23</sup> The first total synthesis of (–)-laulimalide was accomplished in 2000 by Ghosh and Wang,<sup>17a</sup> who later refined their approach by an improved introduction of the 2,3-*cis*-enoate.<sup>17b,c</sup> Almost 1 year later, Ghosh's first report was followed in close succession by three approaches from Mulzer's team,<sup>18a-c</sup> among which one was totally stereoselective, and one from Paterson's group,<sup>19</sup> which before had achieved the synthesis of the fully functionalized macrocyclic core of **1**.<sup>15</sup>

From these early syntheses the following conclusions concerning a successful "endgame" can be drawn.

The macrocycle of **1** cannot be obtained through RCM of a 19-acryloyl 5-allyl-substituted *seco* compound (Figure 2, eq 1).<sup>17c</sup>

Macrocyclization via intramolecular Horner–Wadsworth–Emmons–(HWE) olefination of an aldehyde– phosphonate produces unfavorable 2,3-*E/Z*-mixtures (ca. 2:1) in favor of the (*E*)-isomer (Figure 2, eq 3).<sup>17a,18a</sup>

Base-induced macrolactonization of a 19-hydroxy 2,3-alkenoic acid leads to extensive isomerization of the 2,3-(Z)-enoate (Figure 2, eq 2).<sup>17b,c,18c</sup> However, no E/Z-isomerization is observed during partial hydrogenation of a 2,3-alkynoate macrocycle<sup>17b</sup> or during Mitsunobu-type macrolactonization of a *seco* acid with (19*R*) hydroxy group.<sup>19</sup>

The sensitive epoxide can introduce regio- and stereoselectively in the last step by Sharpless epoxidation of 16,17-allylic alcohol of the unprotected macrocycle (Figure 2, eq 4).<sup>18a,19</sup>

This knowledge proved helpful in the total syntheses by Wender,<sup>20</sup> Crimmins,<sup>21</sup> Williams,<sup>22</sup> and Nelson,<sup>23</sup> which followed in 2002.

#### III. Early Synthetic Work

#### A. First Laulimalide Fragment (Ghosh)

In the first laulimalide-directed communication,<sup>11a</sup> Ghosh's group reported the enantioselective synthesis of methyl ketone **5** as a  $C_3-C_{14}$  segment of **1** which, following the retrosynthetic concept in Figure 3, was to be assembled with an appropriately functionalized  $C_{15}-C_{20}$  epoxyaldehyde **4** by an aldol-type reaction.



**Figure 3.** Ghosh's first retrosynthetic plan: fragment connection between  $C_{14}$  and  $C_{15}$ .

In the event, a hetero Diels-Alder reaction of benzyloxy acetaldehyde and Danishefsky's diene 6 catalyzed by the chiral Cu(II)-bisoxazoline complex 7<sup>25</sup> was used to construct dihydropyranone 8 enantioselectively (62% yield, ee = 85%). After conversion of 8 to acetate 9, the  $C_3-C_4$  side chain was transstereoselectively appended by a Ferrier-type reaction with vinyl-OTBS using montmorillonite K10 clay as a Lewis acid.<sup>26</sup> To install the  $C_{11}$  methyl group, aldehyde 10 was converted in nine steps to Nenoylsultam 11, which by reaction with Me<sub>2</sub>CuLi afforded diastereoselectively conjugate addition product 12, albeit with low conversion and in unsatisfactory yield. Alkylation product **12** was then transformed to methyl ketone 5 via the corresponding acid<sup>27</sup> (Scheme 1).

#### B. Fragment Syntheses by Nishiyama

The synthetic efforts of Nishiyama and Shimizu<sup>12</sup> did not, until now, lead to a total synthesis of **1**. In their retrosynthetic analysis (Figure 4), the Japanese group planned the convergent assembly of the laulimalide skeleton by an allylation of the chiral amide **15** with allyl iodide **14**, controlling the stereochemistry at  $C_{11}$  by the Evans oxazolidinone protocol.<sup>28</sup> The

Scheme 1. Ghosh's First Laulimalide Fragment: Synthesis of Methyl Ketone 5



 $C_{12}-C_{27}$  fragment, in turn, was to be assembled from subunits **16**, **17**, and **18**.

In their first approach,<sup>12a</sup> the bond construction between  $C_{10}$  and  $C_{11}$  was tested successfully with a  $C_{12}-C_{16}$  model iodide and served to synthesize a  $C_{1-}$   $C_{16}$  fragment of **1** (Schemes 2 and 3). The synthesis started from D-mannose pentaacetate (**19**), which was



Figure 4. Nishiyama's retrosynthetic analysis.





Scheme 3. Syntheses of Allyl Iodide 27 and  $C_1-C_{16}$ Fragment 30 by Nishiyama and Shimizu



transformed to the known tetrahydropyran **21** by Nicolaou's six-step procedure.<sup>29</sup> Compound **21** was converted to aldehyde **22** by routine operations. Olefination of **22** with the chiral phosphonate **23** led selectively to (*E*)-enamide **24** in moderate yield. The

double bond in **24** was hydrogenated in the presence of a Wilkinson catalyst, and after selective removal of the isopropylidene group, the resulting diol was deoxygenated to key compound **15**.

The synthesis of allyl iodide 27 (Scheme 3), serving as a model compound in the critical allylation step, began with the L-glutamic acid-derived diester 25,<sup>30</sup> which was selectively reduced at the  $C_{16}$  position. After protection of the resulting diol, reduction of the  $C_{12}$  ester group followed by oxidation afforded aldehyde **26**. Eschenmoser methylenation<sup>31</sup> of aldehyde 26 led to an enal, which was smoothly converted to the desired iodide **27**. The lithium enolate of **15** was then alkylated with iodide 27 to provide amide 28 with the required configuration at C<sub>11</sub>. Reduction of the carboxamide to the methyl group was accomplished in three steps and led to the 11-methyl derivative **29** in good yield. After selective cleavage of the MOM ether in 29 and Parikh-Doering oxidation of the alcohol,<sup>32</sup> the resulting aldehyde was subjected to a Wittig reaction under Mukaiyama-Suzuki conditions,<sup>33</sup> leading to 2,3-(Z)-enoate **30** in rather low yield.

Following the retrosynthetic plan in Figure 4, Nishiyama and Shimizu reported the synthesis of the extended allyl iodide 14 with the complete  $C_{12}-C_{27}$ moiety (Scheme 4).<sup>12b</sup> The E double bonds in 14 were to be generated by an HWE olefination of aldehyde **16** with  $\beta$ -oxophosphonate **17** (C<sub>21</sub>=C<sub>22</sub>) and by a classical Julia olefination between a C<sub>17</sub> phenyl sulfone and aldehyde **18** ( $C_{16}=C_{17}$ ), respectively. Aldehyde 18, which represents the  $C_{12}$ - $C_{16}$  part, was again prepared from diester 25,30 which was converted to PMP-acetal **31** in four steps. α-Methylenation<sup>31</sup> of aldehyde **31**, followed by reduction and silvlation, led to intermediate 32. Reductive cleavage of the cyclic acetal in 32 proceeded with moderate selectivity, leading to a 5:1 mixture of regioisomers. The primary alcohol was then oxidized to aldehyde 18.

The exocyclic dihydropyran fragment **16** was prepared according to the Jørgensen protocol,<sup>34</sup> by asymmetric hetero-Diels—Alder (HDA) reaction of isoprene and ethyl glyoxylate in the presence of (R)-(+)-BINOL-Al-Me as catalyst and ensuing ester to aldehyde interconversion (no further details were given in the communication). This cycloaddition is known to proceed with high enantiocontrol (up to 97% ee), but the HDA adduct is formed as a 2:1 mixture with the corresponding ene product.<sup>34</sup> (Later on, the HDA reaction between isoprene and methyl glyoxylate reappeared in the total synthesis of Wender,<sup>20</sup> who utilized Mikami's catalyst<sup>35</sup> instead; see Scheme 31.)

The synthesis of the  $C_{17}-C_{20}$  subunit (Scheme 4) started with natural (*S*)-malic acid (**33**), which was transformed to methyl ester **34** (no preparation given). Acylation of dimethyl (lithiomethyl)phosphonate with ester **34** afforded  $\beta$ -oxophosphonate **17**, which was connected with aldehyde **16** to provide selectively (*E*)-enone **35** in moderate yield. Diastereoselective carbonyl reduction of **35** with L-selectride furnished the desired 20*S*-configuration but evidently led to extensive silyl migration. This would explain the following transformations leading to isopropy-





lidene acetal **36**, which resulted in the loss of orthogonal protective groups at the vicinal  $C_{19}$  and  $C_{20}$  hydroxy groups. Deprotection of the PMB ether in **36** and phenylsulfonylation of the resulting primary alcohol via the mesylate provided sulfone **37**. The subsequent three-step Julia olefination between **37** and aldehyde **18** led to (*E*)-olefin **38** as the sole product; however, the overall yield of the coupling was unsatisfactory (32% for three steps).<sup>36</sup> To complete the synthesis of key fragment **14**, the silyl ether in coupling product **38** was cleaved and the resulting primary alcohol halogenated via the mesylate.

# C. First RCM-Based Fragment Syntheses (Mulzer and Ghosh)

In September 1999, Mulzer and Hanbauer reported the first ring-closing metathesis  $(RCM)^{24}$  approach to the crucial dihydropyran ring in a  $C_1-C_{12}$  fragment of laulimalide (Scheme 5).<sup>13a</sup> The synthesis

Scheme 5. Mulzer's RCM-Based Route to a  $C_1\!-\!C_{12}$  Fragment



started from commercially available methyl (S)-2methyl-3-hydroxypropionate (39), which was converted to the known alcohol **40**<sup>37</sup> and then homologated to aldehyde **41** via the corresponding cyanide. The  $C_9$  stereocenter was efficiently installed by asymmetric Brown allylation<sup>38a</sup> under "salt-free conditions" at -100 °C,<sup>38b</sup> which led to the homoallylic alcohol 42 in excellent yield. Transacetalization of 42 with acrolein diethylacetal furnished the mixed acetals 43 in 84% yield. This reaction was carried out under azeotropical removal of ethanol by an improved modification of Crimmins' protocol.<sup>39</sup> RCM of dienes 43 with Grubbs' first-generation Ru catalyst proceeded smoothly to give an anomeric mixture of ethyl glycosides 44 in 94% yield.<sup>40</sup> The C<sub>5</sub> stereocenter was generated by Lewis acid-mediated C-glycosidation of **44** with vinyl–OTBS in the presence of montmorillonite K-10<sup>26</sup> or lithium perchlorate in ethyl acetate<sup>41</sup> to afford aldehyde 45 as a single isomer. Still-Gennari olefination<sup>42</sup> of **45** led to Z-enoate **46**, which was converted to tosylate 47 in four steps. In the end, tosylate 47 was not used in Mulzer's subsequent total syntheses, in contrast to aldehyde 45 (cf. Schemes 12 and 27).

Independently, Ghosh and Wang reported the synthesis of an advanced  $C_2-C_{16}$  intermediate of **1**, which features a slightly different RCM strategy to elaborate the dihydropyran ring and a Julia olefina-





tion sequence for introduction of the C<sub>13</sub> exo-methylene unit (Scheme 6).<sup>11b</sup> The synthesis began with the conversion of methyl (S)-2-methyl-3-hydroxypropionate (39) to homoallylic alcohol 48, which is the benzyl-protected analogue of Mulzer's intermediate 42. Alcohol 48 was acylated with acryloyl chloride, and acrylate 49 was exposed to Grubbs' first-generation Ru catalyst (10 mol %) in the presence of Ti-(OiPr)<sub>4</sub> (30 mol %) to provide lactone **50**.<sup>43</sup> To introduce the side chain at  $C_5$ , lactone **50** was reduced and the lactol acetylated in situ. Exposure of the resulting acetate to allyltrimethylsilane in the presence of BF<sub>3</sub>. OEt<sub>2</sub> furnished product **51** as a single isomer, which was transformed into iodide 52 in two steps. For the simultaneous installation of the C<sub>13</sub> methylene unit and the  $C_{15}$  hydroxyl group, the (*R*)-glycidol-derived PMB ether 53 was treated with the sodium enolate of methyl phenylsulfonyl acetate to generate a 2.4:1 mixture of  $\alpha$ -phenylsulfonyllactones 54, which was deprotonated and alkylated with iodide 52. Alkylation product 55, obtained as a single diastereomer in 91% yield, was reduced with LiBH<sub>4</sub> to provide 1,4diol **56**. Perbenzoylation of **56**, followed by treatment of the resulting dibenzoate with magnesium amalgam in ethanol,<sup>44</sup> led to the  $C_{13}$ -methylene derivative **57** with concomitant loss of the  $C_{15}$  benzoate.

In subsequent work, a  $C_{16}$ -aldehyde derived from **57** was connected with a  $C_{17}-C_{27}$  phenyl sulfone by standard Julia coupling and served to investigate RCM methodology as the macrocyclization step in the total synthesis of **1**. However, this strategy was not successful (see Figure 2, eq 3).<sup>17c</sup> A slightly modified analogue of **57**, however, was used in the first total synthesis of laulimalide.<sup>17a</sup>

#### IV. Total Syntheses of Laulimalide and Analogues

# A. First Total Synthesis of Laulimalide (Ghosh and Wang)

Ghosh's alternative and finally successful retrosynthetic plan (Figure 5)<sup>17a,c</sup> involved a convergent



**Figure 5.** Ghosh's improved retrosynthetic analysis: Fragment assembly by Julia olefination.

assembly of the  $C_3-C_{16}$  aldehyde **60** and the  $C_{17} C_{27}$  phenyl sulfone 59 by Julia olefination. The macrocycle was to be obtained either by macrolactonization of a 2,3-Z seco acid (Figure 2, eq 3) or by an intramolecular HWE reaction between a  $C_{19}$ phosphonoacetate and a  $C_3$  aldehyde (Figure 2, eq 2), derived from main fragment 58. The sensitive epoxide would be introduced in the penultimate step by applying the Sharpless protocol<sup>45</sup> to the  $C_{20}$ -OPMB-protected macrocycle. Fragment 60 was to be prepared by slight modification of the sequence outlined in Scheme 6, while coupling partner 59 should come from the addition of an alkynyl anion to a C<sub>20</sub> aldehyde. The dihydropyran units of both key fragments should be generated by ring-closing olefin metathesis.24

Synthesis of the C3–C16 Fragment. In modification of the sequence in Scheme 6, lactone 50 was converted to iodide 62 in six steps (Scheme 7). Alkylation of  $\alpha$ -phenylsulfonyl-lactone 54 with iodide 62 led to intermediate 63 as a 4.2:1 mixture of diastereomers. The C<sub>13</sub> *exo*-methylene group was elaborated by a slightly modified three-step sequence leading to intermediate 64 in 72% overall yield. Protective group manipulations and Swern oxidation completed the synthesis of key aldehyde 60. Scheme 7. Synthesis of  $C_3-C_{16}$  Aldehyde 60 via Sulfone Alkylation and Julia Olefination



Synthesis of the C27-C17 Sulfone 59. To obtain the external dihydropyran subunit, Ghosh and Wang relied again on a RCM strategy (Scheme 8).<sup>11c,46</sup> Thus, copper(I)-catalyzed opening of glycidyl ether 65 with isopropenylmagnesium bromide followed by allylation of the resulting homoallylic alcohol provided diene 66 in excellent yield. The trisubstituted double bond was smoothly formed with Grubbs' first-generation Ru catalyst in dichloromethane at room temperature to give dihydropyran 67. Deprotection, followed by Swern oxidation and Corey-Fuchs homologation,<sup>47</sup> led to dibromo olefin **68**. To complete the synthesis of key fragment 59, glycidyl ether 65 which had already served to obtain enantiopure subunit 68, was also used to install the C<sub>19</sub> stereocenter. Thus, treatment of the lithium salt derived from methyl phenyl sulfone with epoxide 65 furnished an alcohol that was protected as PMB ether 69 and was then converted to aldehyde 70 by deprotection and Swern oxidation. Now the stage was set for coupling **70** with the alkynyl anion derived from precursor 68. Treatment of dibromo olefin **68** with *n*-BuLi followed by reaction of the resulting alkynyl anion with aldehyde **70** proceeded with low stereoselectivity (syn:anti = 1.8:1) in 64% yield. The mixture of the  $C_{20}$  epimeric alcohols 71 was oxidized with Dess-Martin periodinane (DMP) to give an alkynyl ketone, which was reduced with L-selectride to deliver *syn-***71** as a single diastereomer. Selective reduction of the triple bond in *syn*-**71** with Red-Al secured the  $C_{21}-C_{22}$  *E* geometry and furnished allylic alcohol 72-a regioisomer of target compound 59-in 81% yield. To shift the  $C_{19}$ -OPMB group to the vicinal hydroxy group, the PMB ether in 72 was removed with TFA and the resulting diol was transformed to 4-methoxybenzylidene acetal 73, which was regioselectively reduced with DIBALH to the desired  $C_{20}$ -OPMB ether **59** in 52% overall yield from regioisomer 72. The observed







Scheme 9. Fragment Coupling and Completion of

regioselectivity of this reduction is postulated to arise from stabilization of Al chelation by the sulfone oxygens.

**Fragment Coupling and Completion of the Synthesis of 1.** For the crucial Julia coupling,  $\gamma$ -hydroxy-sulfone **59** was lithiated and the resulting dianion treated with aldehyde **60**. Peracetylation of the intermediate  $\beta$ , $\gamma$ -dihydroxysulfones followed by exposure to sodium amalgam furnished a 3.4:1 mixture of olefination products, from which the desired isomer **58** was separated in 34% yield (Scheme 9).

Starting from **58** and several close analogues, attempts were made to construct the macrocycle with  $C_2-C_3 Z$  geometry: RCM of a  $C_{19}-O$ -acryloyl 5-allyl-substituted analogue led to decomposition (Figure 2, eq 1),<sup>17c</sup> and Yamaguchi macrocyclization of a 2,3-*Z*-19-hydroxy *seco* acid was accompanied by extensive *Z*/*E* isomerization (*Z*:*E*  $\approx$  1:2).<sup>17b,48</sup> Finally, encouraged by the successful installation of a 2,3-*Z*-enoate in the macrolactone moiety of phorboxazole,<sup>50</sup> an

intramolecular HWE olefination by Still-Gennari's protocol<sup>42</sup> was carried out to close the 2,3-double bond. Toward this aim (Scheme 9), 58 was acylated with bis(2,2,2-trifluoroethyl)phosphonoacetic acid under Yamaguchi conditions.<sup>51</sup> Selective removal of the TBS group with aqueous acetic acid in THF and oxidation with Dess-Martin periodinane furnished cyclization precursor 74 in 79% overall yield. However, treatment of 74 with K<sub>2</sub>CO<sub>3</sub> in the presence of 18-crown-6 in toluene at -20 to 0 °C afforded an unfavorable 1:2 mixture of Z-75 and E-75 in 84% combined yields. Attempts to improve the amount of Z isomer by changing the protecting group at  $C_{20}$ -O or by using Ando's variant52 of the HWE reaction were not successful. Finally, the overall yield of Z-75 was improved from 28% to 47% by UV irradiation of *E*-**75** in Et<sub>2</sub>O for 50 min. This procedure, however, led to partial decomposition and produced a 1:1 mixture of isomers 75 in 66% yield. The fully protected macrolactone Z-75 was then converted to 1 in three steps. Thus, selective cleavage of the  $C_{15}$ –O– MOM group with PPTS in *tert*-butyl alcohol at 84 °C (45% yield), followed by SAE with (+)-diethyl tartrate,  $^{45}$  and removal of the remaining allylic C<sub>20</sub>– OPMB ether with DDQ (48% yield, two steps) finally provided the first synthetic sample of (–)-laulimalide in 22% overall yield from Z-75.

# B. Stereocontrolled Introduction of the 2,3-Z-Enoate (Ghosh)

Some months later, Ghosh and Wang<sup>17b,c</sup> came up with a stereocontrolled installation of the 2,3-Z-enoate (Scheme 10). Thus, key fragment **58** was

Scheme 10. Ghosh's Improved Synthesis of Z-75 by Stereoselective Introduction of the 2,3-Z-Enoate



converted to aldehyde **76** by protective group manipulation and oxidation with Dess–Martin periodinane. Aldehyde **76** was transformed to alkynoate **77** via Corey–Fuchs homologation.<sup>47</sup> Removal of C<sub>19</sub>-THP group followed by ester hydrolysis afforded the *seco* hydroxy acid, which was cyclized to macrolide **78** under Yamaguchi's conditions.<sup>51</sup> Lindlar hydrogenation of the triple bond<sup>53</sup> provided macrolactone *Z*-**75** as a single isomer in 94% yield.

### C. Last Step Introduction of the Epoxide

Mulzer<sup>18a</sup> and Paterson<sup>19</sup> disclosed two total syntheses of **1** which featured the same endgame. In view of the easy isomerization of **1** to **2**, both authors avoided protective group manipulations at the C<sub>20</sub>-OH function after the introduction of the C<sub>16</sub>-C<sub>17</sub> epoxide and applied regio- and stereoselective Sharpless asymmetric epoxidation (SAE)<sup>45</sup> of the unprotected macrocycle **79** as the last step. This strategy was also used in Mulzer's following syntheses<sup>18b,c</sup> and in the syntheses by Wender<sup>20</sup> and Nelson.<sup>23</sup>

#### 1. Mulzer's First Approach

With respect to the retrosynthetic bond disconnections, the approach of Mulzer and Öhler (Figure 6)



Figure 6. Retrosynthetic analysis of Mulzer and Öhler.

is similar to that of Ghosh. An intramolecular Still– Gennari olefination<sup>42</sup> of phosphonate aldehyde **80** (an analogue of Ghosh's advanced synthetic intermediate **74**) was intended for macrocyclization and introduction of the  $C_2-C_3$ -*Z*-enoate. An *E*-selective one-step Julia–Kocienski olefination<sup>54</sup> was envisioned for connecting main fragments **60** and **81**, and to simplify the preparation of the deprotected macrocycle **79**,  $C_{17}-C_{27}$  sulfone **81** and  $C_3-C_{16}$  aldehyde **60** were MOM-protected at the  $C_{20}$  and  $C_{15}$  allylic alcohols. Both dihydropyran rings of **1** were prepared not only by RCM, but also by other methods. Inexpensive compounds from the chiral carbon pool, derived from D-mannitol, D-glucose, and *S*-malic acid, served to procure the subunits **16**, **82**, and **83**.

Synthesis of the C3–C16 Fragment. The synthesis of the  $C_3-C_{16}$  aldehyde **60**, which also served as a key intermediate in Ghosh's synthesis of **1**, started from the known  $\alpha,\beta$ -unsaturated lactone **85** (Scheme 11), available from tri-*O*-acetyl-D-glucal (**84**) in four high-yielding steps<sup>55</sup> and providing carbons  $C_8-C_{13}$  of the laulimalide skeleton. After TBS protection, conjugate *trans*-addition of Me<sub>2</sub>CuLi led to methyl derivative **86** as a single diastereomer with correct configuration at  $C_{11}$ . To allow inversion at  $C_9$  and introduction of the  $C_{13}$ -phenylsulfonyl group



Scheme 11. Two Syntheses of Ethyl Glycoside 90

lactone, 86 was reduced and the primary hydroxy group of the resulting 1,5-diol selectively protected as the phenylsulfide 87. Mesylation followed by desilylation and ring closure under inversion at  $C_9$ furnished epoxide 88, from which the dihydropyran ring was elaborated in comparable overall yields either by Ghosez's one-pot lactonization<sup>56</sup> or by the well-proven RCM protocol<sup>13a</sup> (cf. Scheme 5). Thus, borontrifluoride etherate-mediated addition of the lithium salt derived from methyl 3-phenylsulfonylorthopropionate<sup>57</sup> to epoxide **88** followed by acidcatalyzed cyclization and base-induced elimination of phenylsulfinic acid led to lactone 89, which was converted to ethyl glycoside 90. Alternatively, epoxide 88 was also smoothly converted into the RCM precursor **91** and then to dihydropyran **90** by the fourstep sequence shown in Scheme 11.

The conversion of intermediate **90** to key aldehyde **60** is outlined in Scheme 12. Stereoselective *C*glycosidation of **90** with vinyl-OTBS, followed by reduction and TBS protection, provided  $C_3-C_{13}$  subunit **82** in 84% overall yield from **90**. Later,<sup>18d</sup> sulfone **82** was also conveniently prepared from previous intermediate **45** (cf. Scheme 5). Toward this end, aldehyde **45** was converted to iodide **62** in four conventional steps. Treatment of **62** with the anion derived from methyl phenyl sulfone and *n*-BuLi led to sulfone **82** in 85% yield.





Further elaboration of the  $C_3-C_{16}$  fragment involved deprotonation of sulfone **82**, followed by BF<sub>3</sub>. Et<sub>2</sub>O-mediated addition to (*S*)-4-methoxybenzyl glycidyl ether (**53**), and MOM protection, which led to a 1:1 mixture of  $C_{13}$  epimeric sulfones **92**. The  $C_{13}$  *exo*methylene group was introduced by treating the lithium salt from sulfones **92** with the carbenoid intermediate prepared in situ from isopropylmagnesium chloride and diiodomethane following a slight modification of Julia's procedure.<sup>58</sup> Methylenation product **93**, obtained in 75% yield, was then transformed to key aldehyde **60** in two steps.

Novel Synthesis of the C3–C16 Fragment via Chirally Catalyzed Ene Reaction. Very recently, Mulzer and Pitts disclosed a highly efficient route to  $C_3-C_{16}$  fragment **60**.<sup>13e</sup> In the novel approach, the  $C_{15}$ stereocenter was created by a chirally catalyzed ene reaction<sup>59</sup> between olefin **95** and ethyl glyoxylate **94** (Figure 7).

The synthesis (Scheme 13) began with the allylation of oxazolidinone **96** with 3-bromo-2-methyl-



**Figure 7.** Improved construction of  $C_3-C_{16}$  fragment **60** via ene reaction.

propene to provide adduct 97 selectively (97%, de = 95%). After reductive removal of the auxiliary, the resulting alcohol was homologated to aldehyde 98 via the nitrile. Brown allylation<sup>38</sup> of **98** generated a homoallylic alcohol with the desired stereochemistry at C<sub>9</sub>, which was elaborated to key intermediate 95 via the previous protocol.<sup>13a</sup> Interestingly, RCM of triene 99 proceeded to the desired dihydropyran 100 without interference from the  $C_{13}$  methylene group. Stereoselective C-glycosidation of **100** was performed using commercially available trimethyl vinyloxysilane as the nucleophile and montmorillonite K 10 as the Lewis acid activator.<sup>26</sup> The resulting aldehyde was then converted to key compound 95 in two steps. Treatment of 95 with ethyl glyoxylate in the presence of a catalytic amount of  $(\check{S})$ - $\check{B}INOL$ -Ti $Br_2^{59a}$  provided the ene product 101 in 74% yield with excellent stereocontrol (de = 95%). MOM protection and ester reduction furnished alcohol 102.

### Scheme 13. Novel Synthesis of C<sub>3</sub>-C<sub>16</sub> Fragment 102



**Synthesis of the C17–C27 Sulfone 81.** Following the retrosynthetic plan in Figure 6, the carbon

skeleton of  $C_{17}-C_{27}$  fragment **81** was assembled by an *E*-selective HWE olefination of aldehyde **16** with the chiral  $\beta$ -oxophosphonate **83** derived from inexpensive (*S*)-malic acid. Altogether, three approaches to aldehyde **16** were elaborated by Mulzer's team,<sup>13b,d</sup> two of which were based on RCM. In the first approach (Scheme 14), glycidyl ether **53** was opened

### Scheme 14. Synthesis of Aldehyde 16 by a Non-RCM Strategy



with the Li salt of ethyl propiolate in the presence of  $BF_3 \cdot Et_2O$  to give alcohol **103**. Stereoselective conjugate addition of Me<sub>2</sub>CuLi and in situ cyclization of the hydroxy ester furnished lactone **104**. Reduction of **104** to the lactol and in situ removal of the anomeric hydroxy group provided dihydropyran **105**, which was transformed to aldehyde **16** in two steps.

The second, more convenient approach to aldehyde **16** (Scheme 15)<sup>13b</sup> started from glycidyl trityl ether

## Scheme 15. Mulzer's First RCM Approach to Aldehyde 16



**106**, which was opened with isopropenylmagnesium bromide under copper(I) catalysis. The resulting alcohol was allylated to diene **107** in high overall yield. Despite the presence of a *gem*-disubstituted double bond, diene **107** furnished dihydropyran **108** quantitatively on exposure to  $2-3 \mod \%$  of Grubbs' first-generation Ru catalyst under high dilution at room temperature. Deprotection of **108** under non-aqueous conditions and ensuing oxidation led to aldehyde **16**.<sup>46</sup>

In Mulzer's third approach to aldehyde **16** (Scheme 16), <sup>13d</sup> a two-directional synthesis<sup>60</sup> was applied to the known diepoxide **110**, <sup>61</sup> readily available from the D-mannitol-derived tetraol **109**. Compound **110** was transformed into tetraene **111**, which on RCM under high dilution gave bis-dihydropyran **112** in 83% yield. No medium-ring-sized cycloolefins were formed across the central acetonide ring, which under these condi-





tions served as a barrier to crossover metathesis.<sup>62</sup> Deprotection and cleavage of the resulting vicinal diol in low-boiling solvents furnished the volatile aldehyde **16** in high yield and purity.

The orthogonally protected phosphonate **83**, which provided the  $C_{19}$  stereocenter of **1**, was prepared in 86% yield from the (*S*)-malic acid-derived lactone **113**<sup>63</sup> and diethyl methanephosphonate in one pot (Scheme 17).<sup>64</sup> Subsequent olefination<sup>65</sup> of **83** with aldehyde **16** afforded enone **114** *E*-selectively in high



yield. Luche reduction<sup>66</sup> of the C<sub>20</sub> carbonyl group in **114** at -95 °C led to an 8:1 mixture of C<sub>20</sub> epimeric alcohols in favor of the required epimer *syn*-**115** in 99% combined yields. After separation by HPLC, *anti*-**115** was recycled by oxidation and *syn*-**115** was converted to the primary alcohol **116**. Treatment of **116** with 1-phenyl-1*H*-tetrazole-5-thiol (PT-SH) under Mitsunobu conditions<sup>67</sup> followed by oxidation of the thioether led to key fragment **81**, ready for connection to C<sub>3</sub>-C<sub>16</sub> fragment **60**.

**Fragment Assembly and Completion of the Synthesis.** The crucial fragment assembly (Scheme 18) was performed via a one-step Julia–Kocienski

#### Scheme 18. Fragment Union by Julia–Kocienski Olefination and Macrocyclization by Intramolecular Still–Gennari Olefination



olefination.<sup>54</sup> The potassium salt of sulfone **81** was treated with aldehyde **60** to provide olefins *E*-**117** and *Z*-**117** in 62% combined yields as an 11.4:1 mixture, from which the *E* isomer was isolated in 57% yield. When the reaction was performed in THF, the overall yield rose to 71% but the *E*/*Z* ratio dropped to

2.8:1.<sup>18d</sup> The PMB ether in *E*-117 was cleaved, and the resulting alcohol was acylated with bis(2,2,2trifluoroethoxy)phosphonoacetyl chloride<sup>68</sup> to provide phosphonate 118, which was converted to the cyclization precursor 80 by acid-promoted cleavage of the silvl ether and oxidation with Dess-Martin periodinane (Scheme 18). The crucial intramolecular HWE olefination of phosphonate aldehyde 80, performed under Still's optimized conditions<sup>42</sup> and avoiding an excess of base (0.95 equiv of KHMDS, 6 equiv of 18-crown-6, THF, 50 min, -78 °C), disappointingly led to a mixture (Z:E = 1:1.8) of macrocycles **119** and **120** in 80% combined yield. Later,<sup>18d</sup> the olefination was performed with  $K_2CO_3/18$ -crown-6 in toluene at room temperature, conditions that improved the *Z*-selectivity during phorboxazole ring closure.<sup>50</sup><sup>c</sup> This procedure provided the olefination products quantitatively but did not improve the isomer ratio (Z:E =1:2.1). Separation of 119 and 120, followed by simultaneous removal of both MOM groups with dimethylboron bromide,<sup>69</sup> generated the 16,17-deoxylaulimalides 79 and 121 in 96% and 85% yield, respectively.

Exposure of 2,3-Z-isomer **79** to Sharpless' asymmetric epoxidation (SAE)<sup>45</sup> with natural (+)-diisopropyl tartrate (DIPT) for 2 h at -20 °C proceeded with clean epoxidation at the "matched" allylic site to give a 2:1 mixture of (-)-1 and unreacted compound **79**, from which **1** was isolated in 86% yield, based on recovered starting material (Scheme 19, eq 1).

#### 2. Synthesis of Laulimalide Analogues (Mulzer)

In subsequent work,<sup>18d</sup> Mulzer's group disclosed the results of additional epoxidation experiments with the deoxylaulimalides **79** and **121** (Scheme 19, eqs





2-4). When the epoxidation of **79** was repeated under the same conditions, however, without the tartrate additive (Scheme 19, eq 2), **1** was formed selectively, albeit in a distinctly slower reaction. This result

underscores that there is an intrinsic preference for the 16,17-epoxidation, which means that epoxidation with natural tartrate represents the case in which substrate and reagent control are "matched" to each other. To test the "mismatched" case also, SAE of **79** with (–)-DIPT was also investigated (Scheme 19, eq 3). This reaction led to the "unnatural" 21,22-epoxide **122** regio- and stereoselectively, underlining the power of the SAE reaction. Epoxidation of *E*-isomer **121** in the presence of (+)-DIPT under the same conditions furnished the expected 16,17-epoxide **123**, however, in lower yield (Scheme 19, eq 4).

### 3. Ring Closure by Mitsunobu-Lactonization (Paterson's Approach)

In contrast to the previous syntheses, Paterson's plan (Figure 8)<sup>19</sup> aimed for fragment assembling by



Figure 8. Retrosynthetic analysis of Paterson et al.

asymmetric aldol reaction of  $C_1-C_{14}$  methyl ketone **126**, already containing the *Z*-enoate, and  $C_{27}-C_{15}$ aldehyde **125**. The  $C_{19}$  stereocenter in *seco* acid **124** was to be inverted by a Mitsunobu-type<sup>67</sup> macrolactonization. This protocol had been applied successfully in Paterson's previous synthesis of the macrocyclic core of **1**<sup>15</sup> and has served to install the sensitive (*Z*)-enoate without *Z*/*E* isomerization. Aldehyde **125**, in turn, was to be prepared by aldol reaction of aldehyde **16** with C<sub>20</sub> methyl ketone **127**.

**Synthesis of the C1–C14 Fragment 126.** The synthesis of key fragment **126** (Scheme 20) started with the construction of the *trans*-disubstituted dihydropyran unit of **1** by asymmetric boron aldol





methodology. Thus, reaction of the boron enolate derived from  $\beta$ -chlorovinyl methyl ketone (**128**) with aldehyde 129 provided intermediate 130 in 56% yield, installing the  $C_9$  stereocenter with 80% ee. Alcohol 130 was cyclized to the vinylogous lactone 131 in 61% yield. After conversion to acetate 132, the C<sub>5</sub> stereocenter was introduced by a Ferrier-type reaction with vinyl-OTBS to provide aldehyde 133 in 80% yield. The aldehyde, which had already been used for the syntheses of swinholide A<sup>70a,b</sup> and scyptophycin,<sup>70c</sup> was then converted into Z-enoate **134** by Still–Gennari olefination<sup>42</sup> under careful control of the reaction conditions. Homologation of aldehyde 135, obtained in two steps from 134, in a further HWE reaction provided E-enone 136 stereoselectively. 1,4-Addition of Me<sub>2</sub>Zn to 136 in the presence of catalytic Ni(acac)<sub>2</sub><sup>71</sup> generated a separable mixture of adducts 126 and 11-epi-126 with only a slight excess of the desired 11R-isomer **126** (126:11-epi-126 = 1.6:1).

**Synthesis of the C15–C27 Fragment 125.** In contrast to the previous syntheses of Ghosh and Mulzer, Jacobsen's hetero-Diels–Alder (HDA) reaction<sup>72</sup> was used for the enantioselective construction of the exocyclic dihydropyran subunit (Scheme 21). Thus, exposure of a neat mixture of diene **137** and aldehyde **138** in the presence of molecular sieves to preformed chromium(III) Lewis acid catalyst **139** (5 mol %) gave HDA adduct **140** in 91% yield and with 95% ee. Reductive removal of the anomeric methoxy group followed by deprotection and Swern oxidation

# Scheme 21. Synthesis of Aldehyde 16 by HDA Chemistry



furnished aldehyde **16**, which was to be connected with methyl ketone **127** according to the retrosynthetic plan.

The synthesis of methyl ketone **127** (Scheme 22) started from unnatural dimethyl (*R*)-malate, which





was converted to alcohol **142** via diol **141**.<sup>73</sup> The  $C_{16}$ - $C_{17}$  double bond was efficiently introduced by HWE olefination of the aldehyde derived from **142** with trimethyl phosphonoacetate, and (*E*)-enoate **143** was then transformed to methyl ketone **127** by a series of conventional steps. Boron-mediated aldol coupling of **127** with aldehyde **16** followed by base-induced elimination of the adduct via the corresponding mesylate provided (*E*)-enone **145** as a single stereo-isomer. Chelation-controlled reduction of **145** with Zn(BH<sub>4</sub>)<sub>2</sub> selectively produced *anti*-alcohol **146**. After TBS protection, the primary TBDPS ether was cleaved using TBAF buffered with acetic acid. Oxida-

tion of the resulting alcohol with Dess–Martin periodinane completed the synthesis of  $C_{27}-C_{15}$  aldehyde **125**, ready for the connection to the  $C_1-C_{14}$  fragment **126**.

**Fragment Assembly and Completion of the Total Synthesis.** The aldol coupling of fragments **125** and **126**, mediated by (+)-Ipc<sub>2</sub>BCl/Et<sub>3</sub>N,<sup>74</sup> furnished an inseparable 4:1 mixture of alcohols **147** and 15-*epi*-**147** (for an approach with improved stereocontrol, see Scheme 44).<sup>23</sup> Without separation, this mixture was converted in five steps to a mixture of *seco* acid **124** and its C<sub>15</sub> epimer. As the direct hydrolysis of the methyl ester resulted in 2,3-*Z*/*E*isomerization, the three-step reduction/oxidation sequence shown in Scheme 23 was performed to

Scheme 23. Fragment Connection and Completion of the Total Synthesis of 1



produce acids **124** and 15-*epi*-**124**. Mitsunobu-type<sup>67</sup> macrolactonization of this mixture of hydroxy acids provided macrolide **148** without isomerization of the (*Z*)-enoate, along with the now separable  $C_{15}$  epimer. Introduction of the  $C_{13}$  *exo*-methylene via the Takai reagent<sup>75</sup> followed by deprotection of the two allylic TBS ethers led to deoxylaulimalide **79**, which was epoxidized selectively to (-)-**1** by SAE<sup>45</sup> in the presence of (+)-DIPT.

#### D. Chiral Allylsilane Addition

#### 1. For Macrocyclization (Mulzer and Enev)

Soon after, a fully stereocontrolled route to laulimalide was reported by Mulzer and Enev.<sup>18b</sup> Retrosynthetically (Figure 9), the carbon skeleton of deoxylaulimalide **79** was to be assembled from phosphonoacetate **150** and aldehyde **151**, generating the sensitive 2,3-*Z*-enoate now by an *intermolecular* Still–Gennari olefination.<sup>42</sup> In a hitherto unprecedented approach, macrocyclization was to be per-



Figure 9. Retrosynthetic analysis of Mulzer and Enev.

formed by closing *seco* compound **149** via an *intramolecular* allyl transfer, the stereochemistry at  $C_{15}$  being controlled by a chirally substituted acetal.<sup>76–79</sup>

**Synthesis of AllyIsilane 151.** The synthesis of  $C_3-C_{14}$  fragment **151** started from commercially available ethyl hydrogen (*R*)-3-methylglutarate **152** (Scheme 24), which provided the  $C_9-C_{13}$  segment of

#### Scheme 24. Synthesis of C<sub>3</sub>-C<sub>14</sub> Allylsilane 151



1. Acid 152 was transformed to homoallylic alcohol **153** in four steps, the C<sub>9</sub> stereocenter being installed via Brown's asymmetric allylboration.<sup>38</sup> Elaboration of the dihydropyran ring via RCM strategy and stereoselective introduction of the two-carbon side chain at C<sub>5</sub> was performed as before,<sup>13a</sup> leading to amide 154, which was transformed to methyl ketone **155** in high yield. For the introduction of the allylsilane moiety, methyl ketone 155 was converted to the enolate under kinetic control and treated with  $PhNTf_{2}^{80}$  to provide enoltriflate **156** as a single regioisomer, which was coupled with trimethylsilylmagnesium chloride under Stille's conditions<sup>81</sup> to afford allylsilane 157 in excellent yield. Removal of the TES ether with K<sub>2</sub>CO<sub>3</sub>/MeOH followed by oxidation with Dess-Martin periodinane led to key aldehyde 151.

Synthesis of Phosphonoacetate 150. Phosphonate 150 (Scheme 25) was easily derived from a





bond *E*-selectively by olefination with aldehyde **16**. Reduction of the C<sub>20</sub> carbonyl group of enone **162** under Luche conditions<sup>66</sup> followed by MOM protection of the secondary and liberation of the primary alcohol led to propargylic alcohol **163** with the correct stereochemistry at C<sub>20</sub>. Stereoselective reduction of the triple bond in **163** with Red-Al followed by oxidation of the resulting *E*-allylic alcohol with Dess– Martin periodinane and acetalization with commercially available (*R*,*R*)-(+)-pentane-2,4-diol led to intermediate **164**. Removal of the silyl ether in **164** followed by acylation with bis(2,2,2-trifluoroethoxy)phosphinylacetyl chloride<sup>68</sup> completed the synthesis of key fragment **150**, which was needed for the olefination of aldehyde **151**.

**Fragment Assembly and Completion of the Total Synthesis of 1.** To connect fragments **150** and **151** (Scheme 26), phosphonate **150** was deprotonated





with KHMDS in the presence of 18-crown-6 and treated with aldehyde 151 at -78 °C, carefully avoiding an excess of base. Under these conditions, *Z*-enoate **149** with the complete laulimalide skeleton was obtained as a single isomer in 85% yield. The macrocyclization was performed by adding seco compound 149 slowly at -50 °C to a highly diluted solution of EtAlCl<sub>2</sub> in dichloromethane, providing cyclization product 165 with 15S-configuration in 85% yield. The  $\gamma$ -hydroxy ether remaining at C<sub>15</sub>-O was removed by oxidation to ketone 166 and subsequent  $\beta$ -elimination with TsOH in chloroform. The remaining MOM ether in intermediate 167 was cleaved with dimethylboron bromide at -78 °C<sup>69</sup> to provide deoxylaulimalide 79 in high yield. The conversion of 79 to 1 via reagent matched SAE was then performed according to Mulzer's first synthesis.

#### 2. For Fragment Union (Mulzer and Hanbauer)

With respect to the bond formation between  $C_{14}$  and  $C_{15}$ , the synthesis disclosed by Mulzer and Hanbauer<sup>18c</sup> is closely related to that of Mulzer and Enev (Figure 9).<sup>18b</sup> In this approach, a macrolactonization of *seco* acid **168** was planned as the ringclosing step, whereas an *intermolecular* diastereoselective addition of allyl silane **169** to the chiral acetal **164**<sup>76–79</sup> was envisioned to connect the main fragments (Figure 10).



**Figure 10.** Retrosynthetic analysis of Mulzer and Hanbauer.

**Synthesis of Allylsilane 169.** For the synthesis of allylsilane **169** (Scheme 27), the previous inter-





mediate **45** (cf. Scheme 5) was elaborated to cyanide **170** in four high-yielding steps. Treatment of **170** with methyllithium led to methyl ketone **171**, which was converted to allylsilane **169** by analogy to the procedure shown in Scheme 24.

**Synthesis of the C15–C27 Fragment 164.** Acetal **164**, which had also been an advanced intermediate in the synthesis of Enev and Mulzer<sup>18b</sup> (cf. Scheme 25), was now prepared along an improved route, which also compares favorably with other syntheses of  $C_{15}-C_{27}$  fragments (cf. Schemes 22, 32, 36, 40, 43, 46, and 47). The synthesis (Scheme 28)





started from commercially available  $\alpha$ -hydroxybutyrolactone 172, easily derived from natural (S)-malic acid.<sup>82</sup> After TBDPS protection, the lactone was converted to the chiral  $\beta$ -oxophosphonate **173** by an one-pot procedure.<sup>64</sup> Olefination of **173** with aldehyde **16** under Masamune–Roush conditions<sup>65</sup> led to (*E*)enone 174 as a single isomer. Reduction of the carbonyl group in 174 under Luche conditions<sup>66</sup> followed by functional group manipulations led to aldehyde 175 in high overall yield. Notably, in contrast to the C<sub>19</sub>-OPMB-protected analogue **114** (cf. Scheme 17), 1,2-reduction of 174 at -78 °C furnished the desired syn-alcohol exclusively. Aldehyde **175** was homologated by another *E*-selective HWE reaction to afford Weinreb-amide 176, which was smoothly converted to key fragment 164 in two steps.

Fragment Union and Completion of the Syn**thesis.** The crucial coupling of acetal **164** with allylsilane **169**, performed in the presence of TiCl<sub>4</sub>, pretreated with a trace of triethylamine, led to adduct **178** stereoselectively in 65% yield. The  $\beta$ -hydroxy ether at the newly created C<sub>15</sub> stereogenic center in 178 was removed by Dess-Martin oxidation to methyl ketone 179 and subsequent base-induced  $\beta$ -elimination to C<sub>15</sub> alcohol **180**, which was converted to the fully protected intermediate 182 by protective group manipulations. Selective Swern oxidation of the primary TES ether<sup>83</sup> in **182** led to  $C_3$  aldehyde 183, which was subjected to a Z-selective Ando-Horner-Emmons olefination.52,84 Treatment of the resulting Z-enoate 184 with TBAF led to the desired seco acid 168. However, Yamaguchi macrocyclization<sup>51</sup> of **168** was accompanied by extensive Z/Eisomerization of the 2,3-double bond and led to the MOM-protected macrolides 119 and 120 as an isomeric mixture (E:Z = 2.7:1).<sup>85</sup>

#### E. Regioselective Macrolactonization with Unprotected Diol (Wender)

In 2002, a series of laulimalide total syntheses appeared, the first of which was submitted by the group of Wender at Stanford.<sup>20</sup> The retrosynthetic strategy outlined in Figure 11 envisioned the con-

1



Figure 11. Retrosynthetic plan by Wender et al.

junction of the 5-vinyl-substituted allylsilane 187 with C<sub>15</sub>-enal 186 by means of an asymmetric Sakurai reaction,<sup>86</sup> followed by C<sub>3</sub> homologation and macrolactonization. The presence of two identical protective groups in fragment 186 called for regioselective macrolactonization of an alkynoic acid with unprotected hydroxy groups at C19 and C20, and alkynoic acid 185 was expected to provide the 18membered ring of 1 and not the 19-membered one of its regioisomer neolaulimalide (3).<sup>1d</sup> Commercially available isopropylidene tartrate 189 was envisioned to provide the *syn*-diol unit in key fragment **186**. This four-carbon synthon comprising C<sub>18</sub>-C<sub>21</sub> of the laulimalide skeleton was to be extended at both ends by two consecutive Wittig olefinations with phosphonium salts 188 and 190, respectively. With respect to the final steps (selective semi-hydrogenation of a highly unsaturated alkynoic macrolide to generate the sensitive Z-enoate<sup>17b</sup> and asymmetric epoxidation of the reagent matched allylic alcohol in deoxylaulimalide 79),18a,19 Wender's approach followed the precedence of previous syntheses.

**Synthesis of Allylsilane 187.** The stereogenic center at  $C_{11}$  of fragment **187** (Scheme 30) was



derived from commercial methyl (*R*)-citronellate (191), which was converted to aldehyde 193 by a known three-step procedure.<sup>87</sup> HDA of aldehyde 193 with Danishefsky's diene 6, catalyzed by Jacobsen's (S,S)-Cr-Salen catalyst 194,88 "under non standard conditions" yielded, after treatment with acid, pyranone 195 in 87% yield and with satisfactory diastereoselectivity. Conjugate cuprate addition to 195 using Lipshutz's procedure,<sup>89</sup> followed by trapping of the resulting enolate with Comins reagent,<sup>80b</sup> afforded enol triflate 196 (74%, de = 82%), which was reduced under Stille's conditions<sup>90</sup> to yield intermediate **197**. Treatment of C<sub>13</sub> ester 197 with excess TMSCH<sub>2</sub>MgCl in the presence of rigorously dried CeCl<sub>3</sub>,<sup>91</sup> followed by silica gel induced Peterson elimination of the intermediate bis-silylmethyl carbinol-generated allylsilane 187.

**Synthesis of C15–C27 Aldehyde 186.** The synthesis of **186** started with the construction of the exocyclic dihydropyran fragment **188**, which was produced according to Mikami's procedure,<sup>35</sup> by asym-





Scheme 32. Synthesis of Key Aldehyde 186 1. LAH



metric hetero Diels-Alder reaction of isoprene and methyl glyoxylate in the presence of the chiral titanium complex 198 (Scheme 31). This reaction was

Scheme 31. Synthesis of Phosphonium Salt 188



reported to produce HDA product 199 enantioselectively (ee = 97%) but in very low yield due to the preferred formation of ene product 200. Ester 199 by the three-step sequence depicted in Scheme 31. The Wittig reaction of **188** with aldehyde **201**,

derived from tartrate 189 in three steps (Scheme 32),<sup>92</sup> resulted, after deprotection with TBAF, in the

formation of a 4.5:1 mixture of isomers 202 in favor of the undesired Z isomer in 67% combined yield. After separation, Z-202 was isomerized using a novel procedure by irradiation of a benzene solution in the presence of hexabutyl distannane to provide an 11.4:1 mixture in favor of the *E*-isomer. *E*-202 was oxidized and the resulting aldehyde homologated by another Z-selective Wittig reaction with the known phosphonium salt 19093 to intermediate 203 with a double bond between C<sub>17</sub> and C<sub>18</sub>. Global deprotection of **203** and subsequent silvlation afforded the tris-silvlated compound 204, from which selective removal of the primary TBS ether was finally achieved using cerium ammonium nitrate.94 Oxidation of the resulting homoallylic alcohol with Dess-Martin periodinane followed by transposition and concomitant isomerization of the double bond completed the synthesis of enal 186, ready for the union with allylsilane 187.

**Fragment Assembly and Completion of the** Total Synthesis of 1. The uniquely complex Sakurai reaction between aldehyde 186 and allylsilane 187, mediated by Yamamoto's acyloxyborane 205<sup>95</sup> (Scheme 33), resulted in the formation of alcohol 206 with complete control of the  $C_{15}$  stereocenter in excellent 86% yield. After MOM protection, chemo- and regioselective hydroboration of the C<sub>5</sub>-vinyl group in the highly unsaturated substrate, performed in the presence of cyclohexene,<sup>96</sup> followed by oxidation led to C<sub>3</sub>aldehyde 207 in high yield. Homologation of 207 using the Bestmann modification of the Gilbert-Seyferth reaction<sup>97</sup> and acylation of the resultant alkyne afforded alkynoate 208. Desilylation and ester

**Scheme 33. Fragment Conjunction and Completion of the Total Synthesis of 1** 



hydrolysis led to the desired 19,20-dihydroxy alkynoic acid **185**, which under Yamaguchi's conditions<sup>51</sup> afforded the 18-membered macrolide **209** regioselectively in 55% yield. Lindlar hydrogenation in the presence of 1-hexene,<sup>53</sup> as previously performed by Ghosh with a close analogue of **209**,<sup>17b</sup> followed by MOM deprotection with dimethylboron bromide<sup>69</sup> provided deoxylaulimalide **79**, which was selectively epoxidized to **1**.

#### F. Early Incorporation of the Epoxide

#### 1. Crimmins' Approach

Crimmins' laboratory disclosed a highly concise total synthesis of **1** that demonstrated for the first time that the sensitive epoxide could be introduced at an early stage.<sup>21</sup> The retrosynthetic design focused on a (19*R*)-hydroxy acid **210** with preformed epoxide moiety, which should serve as the substrate in a Mitsunobu-type<sup>67</sup> macrolactonization (Figure 12). A diastereoselective addition of a C<sub>1</sub>-C<sub>14</sub> allylstannane **212** to a C<sub>15</sub>-C<sub>27</sub>  $\alpha,\beta$ -epoxyaldehyde **211** was envisioned to join the major fragments. The presence of homoallylic (or latent homoallylic) C-O bonds at C<sub>5</sub>,



Figure 12. Crimmins' retrosynthetic analysis.

 $C_{19}$ , and  $C_{23}$  in **1** led to the strategic decision to rely on the glycolate variant<sup>98</sup> of the Evans asymmetric alkylation<sup>28</sup> to construct both subunits. Additionally and in contrast to previous syntheses of **1**, removal of the two TBS protective groups from the  $C_{15}$  and  $C_{20}$  hydroxyls should be attempted as the final step without affecting the *Z*-enoate *and* the epoxide.

Synthesis of the C1–C14 Fragment 212. The synthesis displayed in Scheme 34 started from (S)citronellal (213),<sup>99</sup> which provided the  $C_9-C_{14}$  part of the laulimalide skeleton. Treatment of 213 with Brown's chiral borane<sup>38</sup> produced homoallylic alcohol 214, which was transformed into acyl oxazolidinone **215** by alkylation with bromoacetic acid followed by acylation of the D-valine-derived oxazolidinone. Alkylation of intermediate **215** with the *Z*-allylic iodide **216** proceeded with high diastereoselectivity (de = 94%) to generate intermediate 217 with the required C<sub>5</sub> stereochemistry. The chiral auxiliary in **217** was removed with LiBH<sub>4</sub><sup>100</sup> to provide a primary alcohol, which was converted to tetraene 218 by Swern oxidation and Wittig methylenation. RCM of 218 with Grubbs' first-generation ruthenium catalyst led to dihydropyran 219 in high yield without affecting the other double bonds. Selective cleavage of the trisubstituted double bond in **219**, followed by  $\alpha$ -methylenation of the intermediate aldehyde,<sup>31</sup> and 1,2reduction of the resulting enal led to allylic alcohol 220, which was transformed to allylstannane 212 via the mesylate.

Synthesis of the C15–C27 Fragment 211. The  $C_{22}-C_{27}$  subunit of 1 was also prepared by taking advantage of the asymmetric glycolate alkylation (Scheme 35). Thus, oxazolidinone 221 was alkylated with 2-methylallyl iodide to install the  $C_{23}$  stereocenter in diene 222 (de = 92%). RCM of 222, followed by reductive removal of the auxiliary, and Swern oxidation of the resulting alcohol provided aldehyde 16.<sup>101</sup>

Asymmetric glycolate alkylation was also utilized to establish the  $C_{19}$  stereocenter (Scheme 36). Alkylation of glycolate **223** with (*E*)-allyl iodide **224** produced intermediate **225** (de = 96%), which was

#### Scheme 34. Synthesis of C<sub>1</sub>-C<sub>14</sub> Allylstannane 212



### Scheme 35. Synthesis of Aldehyde 16 via Glycolate Alkylation/RCM Sequence



converted to the chiral  $\beta$ -oxophosphonate **226** via an intermediate Weinreb amide. The C<sub>21</sub>-C<sub>22</sub> double bond and the C<sub>20</sub> stereocenter were then selectively elaborated by HWE olefination of phosphonate **226** with aldehyde **16**, followed by chelation-controlled 1,2-reduction<sup>102</sup> of the resulting (*E*)-enone. Sequential TBS protection of the secondary hydroxy group in **227** and deprotection of the primary one generated allylic alcohol **228**. SAE of **228** (de = 90%) and Dess-Martin oxidation of the intermediate epoxy alcohol generated key fragment **211** in high yield.





**Fragment Assembly and Completion of the Synthesis of 1.** Trimethylaluminum-mediated addition of allylstannane **212** to epoxy aldehyde **211** (Scheme 37) resulted in a 3:1 mixture of C<sub>15</sub> alcohols,

## Scheme 37. Fragment Assembly and Completion of the Total Synthesis of 1





from which the major Felkin–Anh diastereomer **229** with laulimalide stereochemistry at C<sub>15</sub> was isolated in 72% yield. TBS protection of **229**, followed by removal of both PMB groups, led to diol **230**. Selective oxidation of the C<sub>1</sub> allylic alcohol in **230** with MnO<sub>2</sub> led to partial isomerization of the *Z*-enal (*Z*:*E* = 91:9), which was immediately oxidized to *seco* acid **210**. Macrolactonization under Mitsunobu's condi-

tions<sup>67</sup> generated the TBS-protected macrolide **231** in 46% yield, which upon careful exposure to  $Et_3N$ -HF<sup>103</sup> furnished **1** without affecting the *Z*-enoate and without concomitant isomerization to isolaulimalide **(2)**.

#### 2. Williams' Approach

A synthesis of the TBS-protected laulimalide macrocycle **231** was also achieved by the team of D. R. Williams.<sup>22</sup> As outlined in Figure 13, the retrosyn-



**Figure 13.** Retrosynthetic analysis by D. R. Williams et al.

thetic disconnections and also the main fragments are closely related to Crimmins' synthesis. Key fragments 233 and 234 were to be coupled diastereoselectively. In contrast to the work of Crimmins, the crucial allylation of  $\alpha,\beta$ -epoxyaldehyde **233** (C<sub>19</sub> epimer of Crimmins' main fragment **211**) was to be performed with allylsilane 234. In a novel approach, the  $C_{19}-C_{20}$  syn-diol unit in main fragment **233** should arise from a chelation-controlled addition of alkenylzincate 235 to the (S)-malic acid-derived aldehyde **236**. This implied that the subsequent RCM step for generation of the exocyclic dihydropyran moiety was to be performed with a highly unsaturated intermediate. The acetylenic  $C_1 - C_4$  unit in allylsilane 234 was introduced in one step by a novel allenylstannane Ferrier reaction and allowed the introduction of the sensitive Z-enoate through macrolactonization of 2,3-alkynoic acid 232 and subsequent partial hydrogenation.

**Synthesis of the C1–C14 Fragment 234.** The synthesis of allylsilane **234** began with the introduction of the  $C_{11}$  stereocenter by asymmetric conjugate addition of an allylcopper reagent, formed under

Yamamoto's conditions,<sup>104</sup> to *N*-enoyloxazolidinone **237** (Scheme 38). Subsequent removal of the chiral

#### Scheme 38. Synthesis of Allylsilane 234



auxiliary and ozonolysis led to aldehyde 193, which as in the work of Wender (cf. Scheme 30) was elaborated to dihydropyranone 195 by asymmetric HDA with diene **6** in the presence of Jacobsen's (*S*,*S*)-Cr-salen catalyst 194.88 The cycloaddition performed in *t*-BuOMe at -25 °C led, after treatment with acid, to an inseparable mixture of C<sub>9</sub> diastereomers (7.5:1) in 91% combined yield, which was used in the ensuing steps. 1,2-Reduction under Luche's conditions<sup>66</sup> and acetylation provided acetate 238, which was treated with allenylstannane 239 in the presence of BF<sub>3</sub>·Et<sub>2</sub>O to generate the propargylated compound **240** with the required configuration at C<sub>5</sub>. Allenylstannane 239 was prepared from 2-butyn-1,4-diol by PMB monoprotection followed by mesylation and S<sub>N</sub>2 displacement with (Bu<sub>3</sub>Sn)<sub>2</sub>CuLi (eq 1 in Scheme 38, no further details were given in the paper).<sup>105</sup> The preparation of  $C_1 - C_{14}$  fragment **234** was then completed as in Wender's work through cerium chloridemediated double addition of TMSCH<sub>2</sub>MgCl,<sup>91</sup> which resulted in spontaneous elimination to generate the allylsilane moiety.

**Synthesis of Epoxyaldehyde 233.** The synthesis of epoxyaldehyde **233** started with the preparation of alkenylzincate **235** (Scheme 39). Diene **66** (cf. Scheme 8), available in two steps from THP-protected (*R*)-glycidol, was transformed to alkyne **241** by conversion to the corresponding aldehyde and homologation via Bestmann's modification of the Gilbert–Seyferth reaction.<sup>97</sup> Hydrozirconation of alkyne **241** and in situ transmetalation with dimethylzinc, as described by Wipf,<sup>106</sup> yielded the (*E*)-alkenyl derivative **235**, ready for the *syn*-selective coupling with the C<sub>15</sub>–C<sub>20</sub> subunit.





The synthesis of the coupling partner **236** started from natural (*S*)-malic acid (**33**), which was transformed via a known six-step sequence<sup>107</sup> to acetal **242** and then to the  $C_{19}$ -OPMB-protected aldehyde **236** by a series of protective group manipulations (Scheme 40). Aldehyde **236** was treated with alkenylzinc





species **235** to provide a 4:1 mixture of  $C_{20}$  diastereomers in favor of the required (20.5)-alcohol **243**. After TBS protection, a late-stage RCM with Grubbs' first-generation ruthenium catalyst led to the selective formation of the dihydropyran ring, without affecting the other double bonds, albeit in low yield. At this point, the  $C_{20}$  epimers were separated and intermediate **244** converted to epoxyaldehyde **233** by selective cleavage of the primary silyl ether followed by SAE<sup>25</sup> and oxidation with Dess–Martin periodinane.

**Fragment Assembly and Completion of the Synthesis.** The final steps of the synthesis are shown in Scheme 41. Borontrifluoride etherate-mediated allylation of epoxyaldehyde **233** with allylsilane **234** led to the desired Felkin–Anh adduct with 15*S*configuration in 53% yield, which was protected as the TBS ether **245**. Removal of the PMB groups at Scheme 41. Fragment Assembly and Synthesis of the TBS-Protected Laulimalide 231



 $C_1$  and  $C_{19}$  led to a diol that was selectively oxidized to *seco* acid **232**. Yamaguchi macrocyclization, followed by selective Lindlar hydrogenation of the resulting 2,3-alkynoic lactone in the presence of 1-hexene,<sup>53</sup> generated the TBS-protected macrolide **231**. Attempts to remove the silyl groups under standard conditions led to Z/E isomerization and other side reactions. However, the successful conversion of **231** to laulimalide had previously been reported by Crimmins.<sup>21</sup>

#### G. Asymmetric Acyl Halide–Aldehyde Cyclocondensation (Nelson)

A remarkable sequence leading from acetaldehyde to (-)-laulimalide in only 23 steps along the longest linear route has been presented by the team of S. G. Nelson.<sup>23</sup> They were mainly interested in **1** as a platform for evaluating their recently developed asymmetric acyl halide-aldehyde cyclocondensation (AAC) methodology<sup>108</sup> and ensuing transformations of the resulting chiral  $\beta$ -lactones for the synthesis of major fragments of 1. AAC-based bond constructions catalyzed by the chiral Al(III)-triamine complex 251 were used to generate the  $C_{19}$  and  $C_{9}$ ,  $C_{11}$  stereocenters in main fragments 247 and 248 (Figure 14) and served also to construct the dihydropyran moiety in fragment **248** by a novel one-pot  $\beta$ -lactone to dihydropyrone interconversion.<sup>109</sup> Similar to Williams' synthesis,<sup>22</sup> the syn-diol arrangement in fragment 247 was to be generated by chelate-controlled addition of an alkenyl-metal intermediate 249 to a  $C_{15}-C_{20}$  aldehyde **250**. With respect to the asymmetric aldol reaction between  $C_{15}$ - $C_{27}$  enal 247 and  $C_1-C_{14}$  methyl ketone **248** and the late-stage introduction of the C<sub>13</sub> exo-methylene group, the synthesis follows (and improves) Paterson's approach.<sup>19</sup> The stereoselective attachment of the entire  $C_1 - C_4$  moiety by Lewis acid-mediated allenylstannane addition to





a glycal intermediate, also performed by Williams,<sup>22</sup> allows a straightforward access to the 2,3-*Z*-enoate. The last-step epoxidation of deprotected deoxylaulimalide **79**, in turn, was to be performed following procedures described by Mulzer<sup>18</sup> and Paterson.<sup>19</sup>

Synthesis of C1-C14 Fragment 248. The construction of methyl ketone 248 (Scheme 42) commenced with (R)-propiolactone 252, derived from asymmetric AAC between acetaldehyde and acetyl bromide. Type A lactone opening with aluminum N,O-dimethylhydroxylamide<sup>110</sup> and ensuing silyl protection led to Weinreb amide 253, which was transformed to 1,3-*syn*- $\beta$ -lactone **254** by sequential amide to aldehyde interconversion and asymmetric AAC homologation (86% yield, de = 94%). The C<sub>11</sub> stereocenter was then installed by cuprate-mediated S<sub>N</sub>2 (type B) opening of lactone 254 to provide carboxylic acid **255**.<sup>111</sup> After acid to aldehyde interconversion, the configuration at C<sub>9</sub> was established also by asymmetric AAC methodology, leading to anti, anti- $\beta$ -lactone **256** (90% yield, de = 84%). According to a previously developed procedure, <sup>109</sup>  $\beta$ -lactone **256** was





transformed to dihydropyrone **258** by reaction with acetaldehyde equivalent **257** (type A lactone opening) and ensuing acid-mediated cyclization—dehydroamination of the intermediate  $\beta$ -ketohydrazone. Diastereoselective carbonyl reduction, followed by acetylation, and installation of the C<sub>13</sub> carbonyl group led to glycal acetate **259**, ready for the introduction of the C<sub>1</sub>–C<sub>4</sub> moiety. Treatment of **259** with allenyl-stannane **260** (no preparation was given for this compound)<sup>105</sup> in the presence of *n*-Bu<sub>3</sub>SnOTf as the Lewis acid activator completed the synthesis of main fragment **248**.<sup>112</sup>

Synthesis of C15–C27 Aldehyde 247. The synthesis of vinyl anion equivalent 249 (Scheme 43) began with Brown allylation<sup>38</sup> of  $\beta$ -tributylstannyl acrolein 261 using the chiral borane 262 to produce homoallylic alcohol 263 in high yield. Alcohol 263 was *O*-allylated, and the resulting triene was subjected to RCM with Schrock's highly sensitive Mo(VI) catalyst 264.<sup>113</sup> This reaction did not affect the stannyl-substituted double bond and led cleanly to dihydropyran 265. Conversion of stannane 265 to the iodide and subsequent transmetalation to the Grignard reagent furnished the desired vinyl anion equivalent 249.

The preparation of coupling partner **250** was initiated by asymmetric AAC of aldehyde **266** to provide propiolactone **267** with the desired configuration at  $C_{19}$ . Amine-mediated (type A) opening of lactone **267** and hydroxyl group protection delivered Weinreb amide **268**, which was homologated to aldehyde **250** via the orthogonally protected triol **269** by routine functional group manipulation. Coupling of aldehyde





**250** with Grignard reagent **249** proceeded with complete chelate control to alcohol **270** with the desired configuration at  $C_{20}$ , which was converted to key aldehyde **247** in three steps.

Fragment Coupling and Completion of the **Synthesis**. The diastereoselective coupling of key fragments 247 and 248 (Scheme 44) was performed by first converting methyl ketone 248 to the chiral boron enolate derived from bromo borane 271,114 followed by treatment with aldehyde **247**. As a major improvement of the analogous step in Paterson's synthesis (cf. Scheme 23), which was performed with (+)-Ipc<sub>2</sub>BCl and furnished an inseparable 4:1 mixture of C<sub>15</sub> epimers,<sup>19</sup> the aldol addition was now mediated with Corey's reagent 271 and led to a 9:1 mixture of alcohol diastereomers, which after TBS protection gave intermediate 272 in 89% yield. Successive deprotection of the PMB ether and the tert-butyl ester in 272 produced seco acid 246. Modified Yamaguchi macrolactonization and Lindlar hydrogenation led to macrolactone 148, which had been an intermediate in Paterson's synthesis (cf. Scheme 23).<sup>19</sup> Therefore, the synthesis was completed accordingly by Takai

### Scheme 44. Fragment Connection and Completion of the Synthesis via Paterson's Intermediate 148



methylenation at  $C_{13}$ ,<sup>75</sup> global deprotection, and selective SAE<sup>18,19</sup> of the resulting desepoxylaulimalide **79**.

#### V. Syntheses of Laulimalide Fragments

#### A. Davidson's Contributions

To date, the Davidson group, who identified laulimalide as a member of the MSAA family,<sup>2a</sup> has presented three syntheses of main fragments of 1.<sup>14</sup> Originally, Davidson suggested the connection of (19*S*)-aldehyde **273** with  $C_1-C_{14}$  fragment **274** via asymmetric allyl transfer (Figure 15).<sup>14a,b</sup> Anticipating Crimmins' approach (Figure 12, Scheme 37),<sup>21</sup> this concept was later modified,<sup>14c</sup> and (19*R*)-epoxyaldehyde **275** was to be combined with allylstannane **274**.

1



Figure 15. Davidson's retrosynthetic analyses.

In the first communication,<sup>14a</sup> (*S*)-citronellal (**213**) was used to provide the  $C_9-C_{14}$  segment of laulimalide (Scheme 45).<sup>115</sup> To this end, **213** was con-



verted to aldehyde 276 in four steps. Aldehyde 276 was subjected to the conditions of Keck's asymmetric HDA reaction<sup>116</sup> with the Danishefsky-type diene 277 to provide dihydropyranone 278 in moderate yield and with low diastereocontrol. Reduction of the carbonyl group in 278 followed by acetylation and Ferrier-type *C*-glycosidation with vinyl-OTBS produced the trans-disubstituted dihydropyran 279. In an improved approach, aldehyde 276 was allylated using Keck's protocol<sup>117</sup> or with better diastereoselection Brown allylation<sup>38</sup> to give homoallylic alcohol 280. Treatment of 280 with methoxyallene in the presence of Pd(OAc)<sub>2</sub> provided diene 281,<sup>40b</sup> which was subjected to RCM. C-Glycosidation of the resulting methyl glycoside provided aldehyde 279. The sensitive 2,3-Z-enoate was then introduced by Still-Gennari olefination<sup>42</sup> of **279** with 2,4-dimethoxybenzyl (DMB) bis(2,2,2-trifluoroethylphosphono)acetate to afford 282. Removal of the silvl ether present

in **282** followed by Swern oxidation and Eschenmoser methylenation<sup>31</sup> produced an enal, which was reduced to alcohol **283**. The allylic alcohol **283** was acetylated and treated with  $Bu_3SnAlEt_2$  applying Trost's methodology<sup>118</sup> to generate the desired allyl-stannane **274** in moderate yield.

In an adjoining paper,<sup>14b</sup> the  $C_{15}-C_{27}$  aldehyde **273** was prepared starting from commercially available  $\beta$ -hydroxy-lactone **284** (Scheme 46). After PMB pro-

#### Scheme 46. Synthesis of the C<sub>15</sub>-C<sub>27</sub> Fragment 273



tection, the lactone was reduced to the lactol, which was directly olefinated to (*E*)-enoate **285**. Protective group manipulations and a Swern oxidation led to  $C_{15}-C_{20}$  aldehyde **287**, which was to be connected with the exocyclic dihydropyran fragment by means of chelation-controlled addition of a vinyl anion.<sup>119,120</sup>

To obtain the required  $C_{22}-C_{27}$  subunit (Scheme 46), diene **107**, prepared from trityl (*S*)-glycidyl ether according to Scheme 15,<sup>13b</sup> was deprotected and oxidized to aldehyde **288**, which on RCM with Grubbs' first-generation ruthenium catalyst furnished dihydropyran aldehyde **16** in moderate yield. Takai iodoolefination<sup>121</sup> of **16** led to a 6:1 mixture of isomers,

from which (*E*)-vinyl iodide **289** was isolated in 48% yield. Lithiation of **289** followed by transmetalation with Me<sub>2</sub>Zn and coupling with **287** stereoselectively led to (20*S*)-alcohol **290**. However, substantial amounts of methylation product **291** were also formed, which could not be separated at this point. After TIPS protection and reductive removal of the benzoyl protecting group, the mixture of primary alcohols was separated and the desired alcohol oxidized to key aldehyde **273**.

In a second-generation synthesis (Scheme 47),<sup>14c</sup> Davidson's group prepared the (19*R*)-epoxyaldehyde

Scheme 47. Synthesis of Epoxy Aldehyde 275



**275** which was required for a final Mitsunobumacrolactonization.<sup>67</sup> The  $C_{21}-C_{22}$  double bond was to be formed via a Julia–Kocienski coupling reaction.<sup>54</sup> The  $C_{19}-C_{20}$  syn-diol unit was now prepared from L-ascorbic acid (**292**), which was converted to known epoxide **294** in seven steps by slight modifications of the known procedure.<sup>122</sup> The epoxide was opened with the lithium salt generated from TBDPS

propargyl ether, and the resulting alcohol was converted to  $C_{21}$ -aldehyde **296** by a series of protective group manipulations.

The desired coupling partner was now obtained from trityl ether 108, prepared by the RCM strategy outlined in Scheme 15.<sup>13b</sup> Removal of the trityl group followed by reaction with 1-phenyl-1H-tetrazol-5-thiol (PT-SH) under Mitsunobu conditions<sup>67</sup> furnished the thioether, which was oxidized with ammonium molybdate/ $H_2O_2$  to provide a 1:1 mixture of the desired sulfone **298** along with epoxide **297**, which resulted from additional attack on the double bond. Epoxy derivative 297 was recycled to the desired dihydropyran **298** by treatment with iodine/Ph<sub>3</sub>P. The onestep olefination<sup>54</sup> between the anion derived from sulfone 298 and aldehyde 296 in DME led to an unfavorable 1:1.3 mixture in favor of the undesired Z isomer. When DME was replaced by DMF, the isomeric ratio was improved to 5:1, and after separation, (E)-isomer 299 was obtained in 68% yield. Selective removal of the TBDPS ether in 299 and reduction of the resulting propargylic alcohol with Red-Al led to an allylic alcohol, which was transformed to epoxy aldehyde 275 by SAE and ensuing Swern oxidation.

# B. Mulzer's (*S*)-Citronellal Based Route to a C<sub>3</sub>-C<sub>14</sub> Fragment

In contrast to Davidson's (*S*)-citronellal-based route to a  $C_1-C_{14}$  fragment of **1** (cf. Scheme 45), Mulzer (Schemes 48 and 49)<sup>13c</sup> and later also Crimmins (Scheme 34)<sup>21</sup> utilized the isopropylidene group in **213** as a protective group by starting their synthesis of allylic laulimalide fragments with the introduction of stereocenter C<sub>9</sub>. In Mulzer's approach, aldehyde **213** was converted to a 1:1 mixture of epoxide diastereomers **300** and 9-*epi*-**300** via Corey's sulfonium ylide addition.<sup>123</sup> Subsequent Jacobsen's HKR in the presence of catalyst **301**<sup>124</sup> led to the formation of diol **302** along with the desired epoxide **300**. Diol **302** was transformed to **300** by a dehydrative cy-

#### Scheme 48. Synthesis of Epoxide 300



### Scheme 49. Mulzer's Synthesis of Allyl Bromide 308



clization under inversion at  $C_9$  providing epoxide **300** in 76% overall yield from **213**.

The conversion of epoxide **300** to dihydropyrone **304** was effected either by Ghosez's sulfone-based  $\ensuremath{\text{procedure}}^{56}$  or by addition of the anion generated from ethyl propiolate and *n*-BuLi at -95 °C, followed by partial hydrogenation and cyclization (Scheme 49).<sup>125</sup> The electron-rich double bond in **304** was then selectively cleaved to provide the C<sub>14</sub> aldehyde, which was subjected to Eschenmoser methylenation<sup>31</sup> to generate enal **305**. Reduction of both carbonyl groups in **305** with DIBALH and sequential treatment with EtOH/TsOH and acetylation led to intermediate 306, ready for the stereoselective introduction of the  $C_2C_3$ appendage. The C-glycosidation, performed with commercially available vinyl-OTMS, led to the corresponding C<sub>3</sub> aldehyde, which was converted to allyl bromide 308 via acetate 307.

# C. Syntheses of Laulimalide Subunits by Lee et al.

In 2001, a Korean group reported several RCMbased approaches to the dihydropyran subunits of  $1.^{16}$  The *trans*-disubstituted C<sub>3</sub>-C<sub>10</sub> dihydropyran moiety was to be prepared by Burke's tandem glycolate Claisen rearrangement-RCM strategy.<sup>126</sup> Toward this end (Scheme 50), propane-1,3-diol (**309**) was

### Scheme 50. Synthesis of *trans*-Disubstituted Dihydropyran 315



converted to allylic alcohol **310** in four conventional steps. SAE of **310** and ensuing mesylation led to intermediate **311**, which on treatment with Zn/NaI provided secondary alcohol **312** with the required  $C_5$  stereochemistry. The alcohol was etherified with sodium bromoacetate, and the resulting acid was converted into the allyl ester **313**. Rearrangement of **313** via the corresponding TMS–enol ether led to acids **314** as a 4:1 mixture of  $C_9$  epimers, which were separated after esterification with diazomethane. The major isomer was then subjected to RCM with Grubbs' first-generation ruthenium catalyst to provide ester **315** as a  $C_3-C_{10}$  fragment of **1**.

Three different RCM routes to the exocyclic dihydropyran fragment **16** were also presented by Lee's group. The first one<sup>16a</sup> utilized Evans' asymmetric hydroxylation<sup>127</sup> to generate the C<sub>23</sub> stereocenter (Scheme 51). Thus, alkenyl carboxylic acid **316**, prepared from diethyl malonate by standard chemistry, was connected with the L-valine-derived oxazolidinone to provide intermediate **317**. Hydroxylation of **317** with oxaziridine **318** generated  $\alpha$ -hydroxyamide **319** stereoselectively. Reductive removal of the auxiliary led to a diol, which was silylated at the primary and allylated at the secondary hydroxyl group to provide cyclization precursor **320**. RCM of diene **320** followed by deprotection led, after 11 steps, to aldehyde **16**.

"To circumvent the cumbersome use of the oxaziridine reagent", two improved approaches to aldehyde **16** were disclosed in the following communication, <sup>16b</sup> which appeared 6 months later. The first one (Scheme 52) utilized Crimmins' glycolate variation of Evans' methodology.<sup>98</sup> Sodium bromoacetate was converted to allyloxyacetic acid, which was connected with the D-phenylalanine-derived oxazolidinone via the mixed pivaloyl anhydride to furnish intermediate **321**. Alkylation of glycolate **321** with methallyliodide led to a diene with the required C<sub>23</sub> stereocenter, which was cyclized by RCM to provide the dihydropyran in moderate yield. Reductive removal of the auxiliary followed by Swern oxidation led to aldehyde **16**.<sup>128</sup>









Eliminating the need of a chiral auxiliary, Lee et al. reported a third approach to aldehyde **16**,<sup>16b</sup> which parallels previous work of Ghosh (Scheme 8) and Mulzer (Scheme 15), by using (R)-glycidol for the introduction of the C<sub>23</sub> stereocenter (Scheme 53). In

Scheme 53. Synthesis of Aldehyde 16 from (*R*)-Glycidol



this route, the TBDPS-protected glycidyl ether was subjected to the same reaction sequence. Interestingly, the epoxide was regioselectively opened without the use of copper salts, and no silyl shifts during epoxide opening and allylation were observed.

#### VI. General Evaluation

After presentation of the complete synthetic work directed to laulimalide, the following short discussion tries to underscore the basic aspects of laulimalide chemistry and also to highlight some of the novel methodology involved.

#### A. Construction of DHP Fragments

RCM seems to be the most favorable route to construct the exocyclic dihydropyran fragment of **1** (Schemes 8, 15, 16, 35,46, and 53), and it was shown that the ring closure can also be performed regiose-lectively with advanced intermediates already containing the  $C_{21}-C_{22}$  (Scheme 43) and moreover also a  $C_{16}-C_{17}$  double bond (Scheme 40). Enantioselective HDA reaction between isoprene and glyoxylates catalyzed by chiral Lewis acids (Scheme 31)<sup>34,35</sup> is, due to concomitant formation of ene products, less efficient. However, HDA methodology was successfully applied by using an 1-alkoxy-substituted diene and Jacobsen's chiral Cr(III) catalyst, followed by reductive displacement of the resulting anomeric alkoxy substituent (Scheme 21).

Starting from chiral C<sub>9</sub> homoallylic alcohols, which were converted to acrylates (Scheme 6) or more favorable to mixed acrolein acetals (Schemes 5, 11, 13, 24, and 45) also, the endocyclic dihydropyran fragment was mostly constructed by RCM. In the approach presented by Crimmins, an advanced tetraene intermediate was utilized for regioselective cyclization (Scheme 34). Alternatively, a propiolate addition-hydrogenation-lactonization sequence (Scheme 49) or Ghosez's sulfone-based method (Schemes 11 and 49) were applied with comparable success. Asymmetric HDA reaction between an aldehyde and a Danishefsky diene in the presence of chiral catalysts (Schemes 1, 30, 38, and 45) leading to chiral dihydropyrones apparently requires extensive screening of catalysts and reaction conditions to obtain satisfactory stereoselectivities. As a novel strategy, chiral  $\beta$ -lactone to dihydropyrone interconversion was applied in Nelson's total synthesis (Scheme 42). The stereocenter in the  $\beta$ -lactone was created by novel asymmetric acyl halide-aldehyde cyclocondensation (AAC) chemistry, which was amply used in Nelson's approach to 1.

The 5,9-*trans*-disubstitution of the  $C_5-C_9$  dihydropyran fragment was generated commonly by Lewis acid-mediated reaction of ethyl glycosides (without rearrangement) or glycal acetates (via Ferrier rearrangement) with vinyl silyl ethers or more recently by a novel C-propargylation with Marshall-type<sup>105</sup> allenyl stannanes (Schemes 38 and 42), which allows the early introduction of the complete  $C_1-C_4$  moiety. Different approaches were presented by Crimmins, who generated first the  $C_5$  stereocenter by asymmetric glycolate allylation and applied RCM to an advanced intermediate (Scheme 34), and also by Wender, who generated the C<sub>5</sub> stereocenter by conjugate addition of vinyl cuprate to a dihydropyrone, followed by reductive removal of the carbonyl group via the corresponding enol triflate (Scheme 30).

Table 1. Antiproliferative Effects of 1, 2, and Paclitaxel in Drug-Sensitive and -Resistant Cells<sup>2a</sup>

	$IC_{50}$ [nM]				
compound	MDA-MB-435 <sup>a</sup>	SK-OV-3 <sup>b</sup>	SKVLB-1 <sup>c</sup>	resistance factor <sup>d</sup>	
laulimalide (1)	$5.74\pm0.58$	$11.53\pm0.53$	$1.210\pm490$	105	
isolaulimalide ( <b>2</b> )	$1.970\pm97$	$2.570 \pm 290$	$2.650 \pm 1.384$	1.03	
paclitaxel	$1.02\pm0.25$	$1.71 \pm 1.07$	>100.000	>58.480	

<sup>*a*</sup> Human breast adenocarcinoma cell line. <sup>*b*</sup> Human ovarian carcinoma cell line. <sup>*c*</sup> Multidrug-resistant subline of SK-OV-3. <sup>*d*</sup> The IC<sub>50</sub> value of the resistant line SKVLB-1 divided by the IC<sub>50</sub> value of the parental line SK–OV-3.

#### B. Connection of Main Fragments

In the early syntheses, the union of major fragments was achieved by formation of a  $C_{16}-C_{17}$  double bond. While the classical three-step variant of the Julia olefination provided poor results (Schemes 4 and 9), Kocienski's one-step modification led to higher yield and improved *E*-selectivity (Scheme 18). With one exception, the ensuing syntheses uniformly utilized bond construction between  $C_{14}$  and  $C_{15}$  with concomitant creation of the C<sub>15</sub> stereocenter to connect the main fragments. This was achieved either by asymmetric boron-aldol reaction (Schemes 23 and 44) or by allylsilane (stannane) chemistry (Schemes 29, 33, 37, and 41). Only in the approach of Mulzer and Enev, the major fragments were connected by a highly stereoselective formation of the 2,3-Z-enoate through Still-Gennari olefination (Scheme 26).

#### C. Macrocyclization

In the early approaches to 1, construction of the laulimalide macrocycle by intramolecular HWE olefination using Still-Gennari's or Ando's method (Schemes 9 and 18) as well as base-induced macrolactonization (Scheme 29) led to extensive loss of 2,3-Z-geometry. In the subsequent syntheses, the ring closure was therefore performed according to Ghosh's precedence, by macrocyclization of a 2,3-ynoic acid and subsequent Lindlar hydrogenation in the presence of 1-hexene (Schemes 41 and 43). In the synthesis of Wender it was additionally shown that this ring closure proceeds regioselectively in the presence of an unprotected 20-hydroxy group (Scheme 33). Alternatively, macrocyclization with retention of 2,3-Z geometry was also achieved under inversion at  $C_{19}$  by Mitsunobu's protocol (Scheme 23), and it was shown that this reaction can also be performed in the presence of the 16,17-epoxide (Scheme 37). A totally different macrocyclization strategy was applied in the total synthesis of Mulzer and Enev, who achieved the macrocyclization by acetal-directed allyl transfer (Scheme 26).

#### D. The Endgame

In Ghosh's early syntheses, the epoxidation was performed on a 20-OPMB-protected macrocycle and final removal of the PMB ether provided laulimalide in moderate yield. To avoid isomerization to the isolaulimalide skeleton and/or loss of 2,3-Z geometry during a final deprotection step, reagent-matched regio- and stereoselective SAE of the unprotected macrocycle was mostly applied (Schemes 19, 23, 26, 33, and 47). Attempts to remove two TBS ethers from the complete laulimalide skeleton by conventional deprotection methods led indeed to decomposition.<sup>22</sup> However, this deprotection was successfully achieved without isomerization by careful exposure to HF– triethylamine (Scheme 37).

#### VII. Antitumor Activity of Laulimalides and Analogues

In the first cytotoxicity tests with laulimalide (1) and isolaulimalide (2), it was shown that 1 is significantly more active against the KB cell line (5 ng/mL) than 2 (>200 ng/mL).<sup>1a</sup> In the adjoining communication,<sup>1b</sup> the inhibition of cell growth was investigated with synthetic laulimalide diacetate and isolaulimalide, using HT-29 (human colon tumor), P388 (murine lymphoma), A549 (human lung tumor), and HL-60 (human promyelocytic leucemia) cells. IC<sub>50</sub> values in the low micromolecular range, 9-14 and 0.5–6  $\mu$ M, were obtained for C<sub>15</sub>–O,C<sub>20</sub>–O-diacetyl-1 and 2, respectively. In Higa's more recent study,<sup>1d</sup> the same cell lines and additionally the MEL28 line were used to determine the activity of unprotected 1 and its minor congener neolaulimalide (3). Very high activity (IC<sub>50</sub> =  $0.01-0.05 \mu$ M) was observed for both compounds in the same assay.

In February 1999, a mechanism-based screening program, aiming for the discovery of new antimicrotubule agents from natural products, identified **1** and **2** as compounds with microtubule-stabilizing activity.<sup>2a</sup> Treatment of A-10 cells (rat aortic smooth muscle cell line, a nontransformed line) with **1** resulted in a dosedependent reorganization of the microtubule network in the cells and in the formation of microtubule bundles and abnormal mitotic spindles. Coincidentally, **1** and **2** induced nuclear convolution and the formation of multiple micronuclei. Incubation of MDA-MB-435 cells with **1** resulted in mitotic arrest and activation of the proteolytic enzymes that accompany apoptotic cell death.

Like paclitaxel, **1** inhibited the cell proliferation of the drug-sensitive cell lines SK-OV-3 (ovarian carcinoma) and MDA-MB-435 (human breast adenocarcinoma), the IC<sub>50</sub> values being between 5 and 12 nM. Isolaulimalide (**2**) was less potent with values in the low  $\mu$ M range (Table 1). Most importantly, both **1** and **2** also inhibited the proliferation of the multidrugresistant SKVLB-1 cell line (a subline of SK-OV-3) that overexpresses the drug efflux pump P-glycoprotein, whereby laulimalide was as much as 100-fold more potent than paclitaxel in the same assay. These data confirm that **1** and **2** are poor substrates for transport by P-glycoprotein, a property that may provide advantages over the taxanes.

Further exciting data were recently communicated by Hamel et al.<sup>2b</sup> It was shown that **1**, while as active

Table 2. Human Ovarian Carcinoma Cells Resistant to Paclitaxel and Epothilones Remain Sensitive to Laulimalide  $(1)^{2b}$ 

	IC50 [nM] (resistance factor) <sup>a</sup>					
compound	1A9 <sup>b</sup>	PTX10 <sup>c</sup>	<b>PTX22</b> <sup>c</sup>	$A8^d$	B10 <sup>e</sup>	A2780/AD10 <sup>f</sup>
laulimalide (1) epothilone A epothilone B paclitaxel	$\begin{array}{c} 3.9 \pm 0.4 \\ 1.7 \pm 0.3 \\ 0.17 \pm 0.08 \\ 1.7 \pm 0.3 \end{array}$	$\begin{array}{c} 6.0 \pm 1 \; (1.5) \\ 18 \pm 7 \; (11) \\ 0.70 \pm 0.4 \; (4.1) \\ 50 \pm 11 \; (29) \end{array}$	$\begin{array}{c} 6.3 \pm 1 \; (1.6) \\ 4.3 \pm 1 \; (2.5) \\ 0.32 \pm 0.2 \; (1.9) \\ 34 \pm 3 \; (20) \end{array}$	$\begin{array}{c} 9.2 \pm 2 \; (2.4) \\ 93 \pm 30 \; (55) \\ 6.4 \pm 4 \; (38) \\ 13 \pm 2 \; (7.6) \end{array}$	$\begin{array}{c} 15 \pm (0.2) \ (3.8) \\ 125 \pm 25 \ (74) \\ 9.0 \ \pm 5 \ (53) \\ 16 \ \pm 4 \ (9.4) \end{array}$	$\begin{array}{c} 31 \pm 0.6 \; (7.9) \\ 16 \pm 0.6 \; (9.4) \\ 2.6 \pm 2 \; (15) \\ 4000 \pm 900 \; (2400) \end{array}$

 $^{a}$  IC<sub>50</sub> value of the resistant line divided by the IC<sub>50</sub> value of the parental line 1A9.  $^{b}$  Parental cell line, a clone of line A2780.  $^{c}$  Paclitaxel-resistant cell lines selected from 1A9.  $^{d}$  Epothilone A-resistant cell line selected from 1A9.  $^{e}$  Epothilone B-resistant cell line selected from 1A9.  $^{f}$  Multidrug-resistant, P-glycoprotein overexpressing cell line, derived from line A2780, selected in the presence of adriamycin.

as paclitaxel, epothilone A, and eleutherobin in promoting the assembly of cold-stable microtubules, was unable to inhibit the binding of [<sup>3</sup>H]-paclitaxel or a fluorescent paclitaxel derivative to tubulin. Moreover, microtubules formed in the presence of 1 and paclitaxel contained approximately equivalent quantities of both drugs. These findings strongly suggest the existence of a drug binding site on microtubules distinct from that occupied by taxoids. (However, to date, it cannot be excluded that **1** binds already to unpolymerized tubulin or to structurally aberrant polymers). Results obtained with paclitaxeland epothilone-resistant cell lines bearing mutated  $\beta$ -tubulin genes further support this conclusion and underline the high biological potential of 1. Although the paclitaxel-resistant mutants (PTX10, PTX22) remain sensitive to the epothilones and the epothiloneresistant mutants (A8 for epothilone A, B1 for epothilone B) remained partially sensitive to paclitaxel, the relative resistance values (i.e., the  $IC_{50}$ value of the resistant line divided by the IC<sub>50</sub> value of the parental line) for laulimalide (1) were the lowest observed in all cases (Table 2).

To date, only two initial studies exist concerning the effect of synthetic laulimalide derivatives on cell growth.<sup>2b,18d</sup> In Hamel's recent article,<sup>2b</sup> deoxylaulimalide (**79**) and synthetic  $1^{17c}$  were compared to paclitaxel for their effects on the growth of human MCF-7 breast cancer cells. In fact, compound **79**, missing the C<sub>16</sub>,C<sub>17</sub>-epoxide, exhibited reduced potency (1/50 of 1), as demonstrated by the IC<sub>50</sub> values of 360, 7.0, and 2.4 nM for **79**, **1**, and paclitaxel, respectively.

In the recent full account on Mulzer's laulimaliderelated work,<sup>18d</sup> laulimalide (1), deoxylaulimalide (79), and the corresponding analogues with 2,3-Eenoate (desepoxy compound **121** and its 16,17-epoxide **123**) were tested for their effects on the proliferation of two drug-sensitive human breast cancer cell lines (MCF-7, MaTu) and two multidrug-resistant human breast tumor lines (NCI/ADR, MaTu/ADR), along with paclitaxel and epothilone B as standards (Table 3). It turned out that **1** is about as active as paclitaxel against the drug-sensitive cells. Unlike paclitaxel, 1 retained its activity against the drug-resistant cell lines, but in all cases it was significantly less active than epothilone B. Compound 121 (2,3-E-enoate, no epoxide) exhibited no activity at all. Deoxylaulimalide (79) and compound 123 (the 16,17-epoxide derived from 2,3-E-macrolide 121) displayed diminished activity (Table 3).

Table 3. Antiproliferative Effects of Laulimalides	1,
79, 121, and 123, Compared with Paclitaxel and	
Epothilone B <sup>18d</sup>	

•				
	IC <sub>50</sub> [nM]			
compound	MCF-7 <sup>a</sup>	NCI/ADR <sup>b</sup>	MaTu <sup>a</sup>	MaTu/ADR <sup>b</sup>
<b>121</b> <sup>c</sup>	ni <sup>f</sup>	ni	ni	ni
<b>79</b> <sup>d</sup>	89	ni	43	170
123 <sup>e</sup>	54	ni	38	250
laulimalide (1)	3.8	36	3.8	6.0
epothilone B	0.59	3.5	0.46	1.2
paclitaxel	3.2	>1000	3.3	600

<sup>*a*</sup> Human breast tumor cell line. <sup>*b*</sup> Human multidrug-resistant breast tumor cell line. <sup>*c*</sup> 2,3-*E*-16,17-Deoxylaulimalide. <sup>*d*</sup> 2,3-*Z*-16,17-Deoxylaulimalide. <sup>*e*</sup> 2,3-*E*-Laulimalide. <sup>*f*</sup> No inhibition measured up to 100 nM.

#### VIII. Conclusion and Prospects

Laulimalide, a new member of the growing family of nontaxane natural compounds with microtubulestabilizing activity that displays antimitotic activity also against paclitaxel- and epothilone-resistant tumor cell lines, has attracted significant attention in both the synthetic organic and the medicinal communities. A variety of total syntheses was accomplished within the last 3 years. Problems arising from the presence of the (*Z*)-enoate and the epoxide, which are both highly sensitive to isomerization, have been overcome by novel macrocyclization strategies and by stereoselective last-step epoxidation. Due to lack of material, the in vivo evaluation of laulimalide and of the few analogues which have been synthesized to date has yet not been possible. However, as short and economical solutions for the construction of main fragments and their conjunction have been developed, it will be possible to design and synthesize further analogues that might be helpful to identify the critical structural features necessary for improved stability and biological activity.

#### IX. Acknowledgments

This review is dedicated to Professor R. W. Hoffmann on occasion of his 70th birthday. For their excellent contributions to the Viennese laulimalide syntheses, we thank Anjum Ahmed, Valentin Enev, Martin Hanbauer, Kate E. Hoegenauer née Dorling, and Mike Pitts. We also thank Hanspeter Kaehlig for helping with the NMR spectra, Sabine Schneider for numerous HPLC separations, and Gerhard Siemeister, Schering AG, Berlin for the antitumor tests of our compounds. Financial support came from the EU (research network HPRN-CT-2000-00018) and the Austrian Science Fund (FWF project P-13941-CHE).

#### X. References

- (a) Corley, D. G.; Herb, R.; Moore, R. E.; Scheuer, P. J.; Paul, V. J. J. Org. Chem. **1988**, 53, 3644.
   (b) Quinoà, E.; Kakou, Y.; Crews, P. J. Org. Chem. **1988**, 53, 3642.
   (c) Jefford, C. W.; Bernardinelli, G.; Tanaka, J.-i.; Higa, T. Tetrahedron Lett. 1996, 37, 159. (d) Tanaka, J.-i.; Higa, T.; Bernardinelli, G.; Jefford, C. W. Chem. Lett. **1996**, 255. (e) Cutignano, A.; Bruno, I.; Bifulco, G.; Casapullo, A.; Debitus, C.; Gomez-Paloma, L.; Riccio, R. Eur. J. Org. Chem. 2001, 775.
- (2)(a) Mooberry, S. L.; Tien, G.; Hernandez, A. H.; Plubrukarn, A.; Davidson, B. S. *Cancer Res.* **1999**, *59*, 653. (b) Pryor, D. E.; O'Brate, A.; Bilcer, G.; Diaz, J. F.; Wang, Yu; Wang, Yo; Kabaki, M.; Jung, M. K.; Andreu, J. M.; Ghosh, A. K.; Giannakakou, P.; Hamel, E. *Biochemistry* **2002**, *41*, 9109.
- (3) For reviews, see: (a) Nicolaou, K. C.; Dai, W.-M.; Guy, R. K. Angew. Chem., Int. Ed. Engl. 1994, 33, 15. (b) Kingston, D. G. I. Chem. Commun. 2001, 867.
- (a) Stachel, S. J.; Biswas, K.; Danishefsky, S. J. *Curr. Pharm. Des.* **2001**, *7*, 1277. (b) He, L.; Orr, G. A.; Horwitz, S. B. *Drug Discovery Today* **2001**, *6*, 1153. (c) Altmann, K.-H. *Curr. Opin. Chem. J. Comp. J. Comp.* **1** (4) Chem. Biol. 2001, 5, 424. (d) Jordan, M. A. Curr. Med. Chem.: Anti-Cancer Agents 2002, 2, 1. (e) Jimenez-Barbero, J.; Amat-Guerri, F.; Snyder, J. P. Curr. Med. Chem.: Anti-Cancer Agents 2002, 2, 91. (f) Myles, D. C. Annu. Rep. Med. Chem. 2002, 37, 125.
- Sarcodictyins: D'Ambrosio, M.; Guerriero, A.; Pietra, F. Helv. (5)Chim. Acta 1987, 70, 2019 and 1988, 71, 964. Eleutherobin: Lindel, T.; Jensen, P. R.; Fenical, W.; Long, B. H.; Casazza, A. M.; Carboni J.; Fairchild, C. R. J. Am. Chem. Soc. **1997**, 119, 8744. Discodermolide: Gunasekera, S. P.; Gunasekera, M.; Longley, R. E.; Schulte, G. K. J. Org. Chem. 1990, 55, 4912. The original structure of discodermolide was revised in *J. Org. Chem.* 1991, 56, 1346. Dictyostatin: Pettit, G. R.; Cichacz, Z. A.; Gaos, F.; Boyd, M. R.; Schmidt, J. M. J. Chem. Soc., Chem. Commun. F.; Boyd, M. R.; Schmidt, J. M. J. Chem. Soc., Chem. Commun.
  1994, 1111. Peloruside A: (a) West, L. M.; Northcote, P. T.; Battershill, C. N. J. Org. Chem. 2000, 65, 445. (b) Hood, K. A.; Bäckström, B. T.; West, L. M.; Northcote, P. T.; Berridge, M. V.; Miller, J. H. Anti-Cancer Drug Des. 2001, 16, 155. (c) Hood, K. A.; West, L. M.; Rouwe, B.; Northcote, P. T.; Berridge, M. V.; Wakefield, St. J.; Miller, J. H. Cancer Res. 2002, 62, 3356.
- (6) For reviews, see: (a) Nicolaou, K. C.; Roschangar, F.; Vourlou-mis, D. Angew. Chem., Int. Ed. Engl. 1998, 37, 2015. (b) Mulzer, J. Monatsh. Chem. 2000, 131, 205. (c) Nicolaou, K. C.; Ritzen, A.; Namoto, K. Chem. Commun. 2001, 1523. (d) Wartmann, M.; Altmann, K.-H. Curr. Med. Chem.: Anti-Cancer Agents 2002, 2 123
- (7) (a) Sato, B.; Muramatsu, H.; Miyauchi, M.; Hori, Y.; Takase, S.; Hino, M.; Hashimoto, S.; Terano, H. *J. Antibiot.* **2000**, *53*, 123. (b) Sato, B.; Nakajima, H.; Hori, Y.; Hino, M.; Hashimoto, S.; Terano, H. J. Antibiot. 2000, 53, 204. (c) Yoshimura, S.; Sato, B.; Kinoshita, T.; Takase, S.; Terano, H. J. Antibiot. 2000, 53, 615. The structure of FR182877 was revised in J. Antibiot. 2002, 55, C1.
- (a) Chen, Z.; Wang, B.; Chen, M. *Huaxue Xuebao* **1988**, *46*, 1201. (b) Kondoh, M.; Usui, T.; Mayumi, T.; Osada, H. *J. Antibiot.* (8)(1) 1998, 51, 801. (c) Roux, D.; Hadi, H. A.; Thoret, S.; Guenard, D.; Thoison, O.; Paies, M.; Sevenet, T. J. Nat. Prod. 2000, 63, 1070
- (a) Wang, Z.; Yang, D.; Mohanakrishnan, A. K.; Fanwick, P. E.; Nampoothiri, P.; Hamel, E.; Cushman, M. *J. Med. Chem.* **2000**, (9)43, 2419. (b) Couladouros, E. A.; Li, T.; Moutsos, V. I.; Pitsinos, E. N.; Soufli, I. C. Bioorg. Med. Chem. Lett. 1999, 9, 2927. (c) Shintani, Y.; Tanaka, T.; Nozaki, Y. Cancer Chemother. Pharmacol. 1997, 40, 513
- (10)(a) Mulzer, J.; Öhler, E.; Enev, V. E. Adv. Synth. Catal. 2002, 344, 573. (b) Crimmins, M. T. Curr. Opin. Drug Discovery Dev. 2002, 5, 944.
- (11) (a) Ghosh, A. K.; Mathivanan, P.; Cappiello, J. Tetrahedron Lett. **1997**, 38, 2427. (b) Ghosh, A. K.; Wang, Y. Tetrahedron Lett. **2000**, 41, 2319. (c) Ghosh, A. K.; Wang, Y. Tetrahedron Lett. 2000, 41, 4705.
- (a) Shimizu, A.; Nishiyama, S. Tetrahedron Lett. 1997, 38, 6011. (12)(b) Shimizu, A.; Nishiyama, S. Synlett 1998, 1209.
- (13)(a) Mulzer, J.; Hanbauer, M. Tetrahedron Lett. 2000, 41, 33. (b) Dorling, E. K.; Öhler, E.; Mulzer, J. Tetrahedron Lett. 2000, 41, 6323. (c) Dorling, E. K.; Öhler, E.; Mantoulidis, M.; Mulzer, J. Synlett **2001**, 1105. (d) Ahmed, A., Öhler, E.; Mulzer, J. Synthesis 2001, 2007. (e) Pitts, M.; Mulzer, J. Tetrahedron Lett. 2002, 43, 8471
- (14) (a) Nadolski, G. T.; Davidson, B. S. Tetrahedron Lett. 2001, 42, 797. (b) Messenger, B. T.; Davidson, B. S. Tetrahedron Lett.

2001, 42, 801. (c) Sivaramakrishnan, A.; Nadolski, G. T.; McAlexander, I. A.; Davidson, B. S. Tetrahedron Lett. 2002, 43, 213

- (15) Paterson, I.; De Savi, C.; Tudge, M. Org. Lett. 2001, 3, 213.
  (16) (a) Lee, H. W.; Jeong, C.-S.; Yoon, S. H.; Lee, I.-Y. C.; Bull. Korean Chem. Soc. 2001, 22, 791. (b) Lee, H. W.; Yoon, S. H.; Lee, I.-Y. C.; Chung, B. Y. Bull. Korean Chem. Soc. 2001, 22, 1172 1179.
- (17) (a) Ghosh, A. K.; Wang, Y. J. Am. Chem. Soc. 2000, 122, 11027.
   (b) Ghosh, A. K.; Wang, Y. Tetrahedron Lett. 2001, 42, 3399. (c) Ghosh, A. K.; Wang, Y.; Kim, J. T. J. Org. Chem. 2001, 66, 8973.
- (a) Mulzer, J.; Öhler, E. *Angew. Chem., Int. Ed.* 2001, *00*, 3842.
  (b) Enev, V. E.; Kählig, H.; Mulzer J. *J. Am. Chem. Soc.* 2001, *123*, 10764. (c) Mulzer, J.; Hanbauer, M. *Tetrahedron Lett.* 2002, *43*, 3381. (d) Ahmed, A.; Hoegenauer, E. K.; Enev, V. E.; Hanbauer, M.; Kählig, H.; Öhler, E.; Mulzer J. *J. Org. Chem.* 2002, *68*, 2026. (18)2003. 68. 3026.
- Paterson, I.; De Savi, C.; Tudge, M. *Org. Lett.* **2001**, *3*, 3149. Wender, P. A.; Hegde, S. G.; Hubbard, R. D.; Zhang, L. *J. Am.* (19)
- (20)*Chem. Soc.* **2002**, *124*, 4956. Crimmins, M. T.; Stanton, M. G.; Allwein, S. P. *J. Am. Chem.*
- (21)Soc. 2002, 124, 5958.
- (22)Williams, D. R.; Mi, L.; Mullins, R. J.; Stites, R. E. Tetrahedron Lett. 2002, 43, 4841.
- (23) Nelson, S. G.; Cheung, W. S.; Kassick, A. J.; Hilfiker, M. A. J. Am. Chem. Soc. 2002, 124, 13654.
- For reviews, see: (a) Grubbs, R. H.; Chang, S. Tetrahedron 1998, (24)54, 4413. (b) Armstrong, S. K. J. Chem. Soc., Perkin Trans 1 1998, 371. (c) Fürstner, A. Angew. Chem., Int. Ed. 2000, 39, 3012. (d) Trinka, T. M.; Grubbs, R. H. Acc. Chem. Res. 2001, 34. 18.
- (a) Ghosh, A. K.; Mathivanan, P.; Cappiello, J. *Tetrahedron Lett.* **1996**, *37*, 3815. (b) Ghosh, A. K.; Mathivanan, P.; Cappiello, J.; Krishnan, K. *Tetrahedron: Asymmetry* **1996**, *7*, 2165. (25)
- Toshima, K.; Miyamoto, N.; Matsuo, G.; Nakata, M.; Matsumura, S. Chem. Commun. 1996, 1379.
- Rubottom, G. M.; Kim, C. *J. Org. Chem.* **1983**, *48*, 1550. Evans, D. A.; Ennis, M. D.; Mathre, D. J. *J. Am. Chem. Soc.*
- (28)**1982**, *104*, 1737.
- Nicolaou, K. C.; Reddy, K. R.; Skokotas, G.; Sato, F.; Xiao, X.; Hwang, C. K. *J. Am. Chem. Soc.* **1993**, *115*, 3558. Ravid, U.; Silverstein, R. M.; Smith, L. R. *Tetrahedron* **1978**, (29)
- (30)34, 1449.
- (31) Schreiber, J.; Maag, H.; Hashimoto, N.; Eschenmoser, A. Angew. *Chem., Int. Ed. Engl.* **1971**, *10*, 330. For a related protocol, see: Paquette, L.; Dyck, B. P. *J. Am. Chem. Soc.* **1998**, *120*, 5953.
- (32) Parikh, J. R.; von Doering, W. E. J. Am. Chem. Soc. 1967, 89, 5505.
- (33)Tabusa, F.; Yamada, T.; Suzuki, K.; Mukaiyama, T. Chem. Lett. 1984, 405.
- (34)Graven, A.; Johannsen, M.; Jørgensen, K. A. Chem. Commun. 1996, 2373.
- Terada, M.; Mikami, K. J. Chem. Soc., Chem. Commun. 1995, (35) 2391
- (36)Later on, similar disappointing results were obtained by Ghosh, who utilized a close analogue of phenyl sulfone **37** to unify the main fragments (cf. Scheme 9).<sup>17a</sup> For more satisfactory results
- with Kocienski's one step variant, see: Scheme 18.<sup>18a</sup> Walkup, R. D.; Boatman, P. D., Jr.; Kane, R. R.; Cunningham, R. T. *Tetrahedron Lett.* **1991**, *32*, 3937. (37)
- (a) Jadhav, P. K.; Bhat, K. S.; Perumal, T.; Brown, H. C. *J. Org. Chem.* **1986**, *51*, 432. (b) Brown, H. C.; Bhat, K. S.; Randad, R. S. *J. Org. Chem.* **1989**, *54*, 1570. (c) Racherla, U. S.; Brown, H. C. *J. Org. Chem.* **1989**, *56*, 401. (38)
- Crimmins, M. T.; King, B. W. J. Am. Chem. Soc. 1998, 120, 9084. (39)(40)For prior syntheses of similarly functionalized dihydropyrans by sequential acetal formation/RCM, see: (a) ref 39. (b) Rutjes, F. P. J. T.; Kooistra, T. M.; Schoemaker, H. E. Synlett 1998, 192. For additional use in laulimalide synthesis, see: Schemes 11, 13, 24, and 45.
- (41) Grieco, P.; Speake, J. D. *Tetrahedron Lett.* **1998**, *39*, 1275
   (42) Still, W. C.; Gennari, C. *Tetrahedron Lett.* **1983**, *24*, 4405.
- (43) For prior syntheses of  $\alpha,\beta$ -unsaturated lactones by RCM, see: (a) Nicolaou, K. C.; Rodriguez, R. M.; Mitchell, H. J.; van Delft, F. L. Angew. Chem., Int. Ed. Engl. 1998, 37, 1874. (b) Ghosh, A. K.; Cappiello, J.; Shin, D. Tetrahedron Lett. 1998, 39, 4651.
   (c) Brown, H. C.; Wetherill, R. B. J. Indian Chem. Soc. 1999, (d) Ghosh, A. K.; Liu, C. *Chem. Commun.* **1999**, 1743.
   (e) Ramachandran, P. V.; Reddy, M. V. R.; Brown, H. C. Tetrahedron Lett. 2000, 41, 583. (f) Ghosh, A. K.; Bilcer, G. Tetrahedron Lett. 2000, 41, 1003. For a discussion on the beneficial effect of  $Ti(OiPr)_4$  on RCM of acrylates, see: (a) Fürstner, A.; Langemann, K. *J. Am. Chem. Soc.* **1997**, *119*, 9130. (b) Fürstner, A.; Dierkes, T. *Org. Lett.* **2000**, *2*, 2463 and references therein.
- (44) Lee, G. H.; Lee, H. K.; Choi, E. B.; Kim, B. T.; Pak, C. S. Tetrahedron Lett. 1995, 36, 5607.
- (a) Rossiter, B. E. Chiral Catalysis. In Asymmetric Synthesis; (45)Morrison, J. D., Ed.; Academic Press: Orlando, FL, 1985; Vol.

5, p 193. (b) Johnson, R. A.; Sharpless, K. B. In Catalytic Asymmetric Synthesis; Ojima, I., Ed.; VCH Publishers: New York, 1993; p 103.

- (46) For closely related approaches to the external dihydropyran fragment, see Schemes 8, 15, 46, 47, and 53.
- (47) Corey, E. J.; Fuchs, P. L. Tetrahedron Lett. 1972, 14, 3769.
- (48) Base-induced C<sub>2,3</sub>-Z/E isomerization during Yamaguchi-type macrolactonization has been previously observed by Roush during a synthesis of verrucarin B49 and was assumed to occur through a reversible Michael addition of nucleophilic reagents to the active ester intermediate.
- (49) Roush, W. R.; Blizzard, T. A. J. Org. Chem. 1984, 49, 4332 and references therein.
- (50) (a) Forsyth, C. J.; Ahmed, F.; Cink, R. D.; Lee, C. S. J. Am. Chem. *Construction of the second state of the secon* Verhoest, P. R.; Schelhaas, M. J. Am. Chem. Soc. 2001, 123, 4834 and 10942.
- (51) Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. Bull. Chem. Soc. Jpn. **1979**, 52, 1989.
- (52) Ando, K. J. Org. Chem. 1999, 64, 8406 and references therein.
- (53) Ho, T. L.; Liu, S. H. Synth. Commun. 1987, 17, 969. (b) For the use of this procedure in phorboxazole synthesis, see: Evans, D. A.; Fitch, D. M.; Smith, T. E.; Cee, V. J. *J. Am. Chem. Soc.* **2000**, 122 10033
- (54) Blakemore, P. R.; Cole, W. J.; Kocienski, P. J.; Morley, A. Synlett 1998, 26. For recent applications of the Julia–Kocienski protocol *J. Org. Chem.* **2000**, *65*, 3738 and references therein. (b) Liu, P.; Jacobsen, E. N. J. Am. Chem. Soc. **2001**, *123*, 10772. (c) Smith, A. B., III; Safonov, I. G.; Corbett, M. C. J. Am. Chem. Soc. **2001**, *123*, 10772. (d) Tachene D.; McGermiere, T.; Ui, M. Soc. 2001, 123, 12426. (d) Takano, D.; Nagamitsu, T.; Ui, H.; Shiomi, K.; Yamaguchi, Y.; Masuma, R.; Kuwajima, I.; Omura, S. Org. Lett. 2001, 3, 2289. (e) Lee, E.; Song, H. Y.; Kang, J. W.; Kim, D.-S.; Jung, C.-K.; Joo, J. M. J. Am. Chem. Soc. 2002, 124, 204. (?) Lauten M. Columpi L. T. Hichert S.; Smith, N. D.; 384. (f) Lautens, M.; Colucci, J. T.; Hiebert, S.; Smith, N. D.; Bouchain, G. Org. Lett. 2002, 4, 1879.
- (55) (a) Corey, E. J.; Pyne, S. G.; Su, W.-g. Tetrahedron Lett. 1983, 24, 4883. (b) Roth, B. D.; Roark, W. H. Tetrahedron Lett. 1988, 29, 1255. See also: (c) Rollin, P.; Sinay, P. *Carbohydr. Res.* **1981**, 98, 139. (d) Lichtenthaler, F. W.; Rönninger, S.; Jarglis, P. Liebigs Ann. Chem. 1989, 1153.
- Carretero, J. C.; Ghosez, L. *Tetrahedron Lett.* **1988**, *29*, 2059. For recent applications of Ghosez's methodology, see: (a) Nico-(56) laou, K. C.; Patron, A. P.; Ajito, K.; Richter, P. K.; Khatuya, H.; Bertinato, P.; Miller, R. A.; Tomaszewski, M. J. *Chem. Eur. J.* **1996**, *2*, 847. (b) Murakami, N.; Wang, W.; Aoki, M.; Tsutsui, Y.; Sugimoto, M.; Kobayashi, M. *Tetrahedron Lett.* **1998**, *39*, 2349. (c) Surivet, J.-P.; Vatèle, J.-M. *Tetrahedron* **1999**, *55*, 13011. (d) Williams, D. R.; Ihle, D. C.; Plummer, S. V. Org. Lett. 2001. 3. 1383.
- (57) De Lombaert, S.; Nemery, I.; Roekens, B.; Carretero, J. C.; Kimmel, T.; Ghosez, L. *Tetrahedron Lett.* **1986**, *27*, 5099. (58) De Lima, C.; Julia, M.; Verpeaux, J.-N. *Synlett* **1992**, 133. For
- recent applications of Julia's methodology, see: (a) Romo, D.; Meyer, S. D.; Johnson, D. D.; Schreiber, S. L. J. Am. Chem. Soc. 1993, 115, 7906. (b) Smith, A. B., III; Doughty, V. A.; Lin, Q.; Zhuang, L.; McBriar, M. D.; Boldi, A. M.; Moser, W. H.; Murase, N.; Nakayama, K.; Sobukama, M. Angew. Chem., Int. Ed. 2001, 40, 191. (c) Smith, A. B., III; Lin, Q.; Doughty, V. A.; Zhuang, L.; McBriar, M. D.; Kerns, Y. K.; Brook, C. S.; Murase, N.; Nakayama, K. Angew. Chem., Int. Ed. 2001, 40, 196.
- (59) (a) Mikami, K.; Terada, M.; Nakai, T. J. Am. Chem. Soc. 1989, *Li11*, 1940. (b) Mikami, K.; Terada, M.; Nakai, T. *J. Am. Chem. Soc.* **1990**, *112*, 3949. (c) Mikami, K.; Terada, M.; Narisawa, S.; Nakai, T. Org. Synth **1993**, 71, 14. (d) Manickam, G.; Sundaray-an, G. Tetrahedron: Asymmetry **1999**, 10, 2913. (e) Quian, C.; an, G. *Tetrahedron. Asymmetry* **1999**, *10*, 2913. (e) Quilan, C.; Wang, L. *Tetrahedron: Asymmetry* **2000**, *11*, 2347. (f) Evans, D. A.; Tregay, P. W.; Burgey, C. S.; Paras, N. A.; Vojkovski, T. J. Am. Chem. Soc. **2000**, *122*, 7936. (g) Hilpert, H. *Tetrahedron* **2001**, *57*, 7675. (h) For a review, see: Dias, L. C. Curr. Org. Chem. **900**, *4*, 205 Chem. 2000, 4, 305.
- (60) Poss, C. S.; Schreiber, S. L. Acc. Chem. Res. 1994, 27, 9. For a review, see: Magnuson, S. R. Tetrahedron 1995, 51, 2167. For some recent examples of two-directional RCM, see: (a) Burke, S. D.; Quinn, K. J.; Chen, V. J. J. Org. Chem. 1998, 63, 8626.
   (b) Lautens, M.; Hughes, G. Angew. Chem., Int. Ed. Engl. 1999, 38, 129.
   (c) Baylon, C.; Heck, M.-P.; Mioskowski, C. J. Org. Chem. 1999, 64, 3354.
   (d) Heck, M.-P.; Baylon, C.; Noian, S. P.; Mioskowski, C. J. Org. Chem. 1999, 64, 3354. Mioskowski, C. Org. Lett. 2001, 3, 1989. (e) Clark, J. S.; Hamelin, O. Angew. Chem., Int. Ed. 2000, 39, 372.
- (61) Le Merrer, Y.; Duréault, A.; Greck, C.; Micas-Languin, D.; Gravier, C.; Depezay, J.-C. *Heterocycles* 1987, 25, 541.
- (62) For a review, see: Maier, M. E. Angew. Chem., Int. Ed. 2000, 39, 2073. For the clean conversion of a bis-homoallylic alcohol derived from diepoxide 110 to the corresponding coloctene derivative by RCM, see: Gravier-Pelletier, C.; Andriuzzi, O.; Le Merrer, Y. Tetrahedron Lett. 2002, 43, 245.

- (63) Mulzer, J.; Mantoulidis, A.; Öhler, E. J. Org. Chem. 2000, 65, 7456
- (64) For related protocols, see: (a) Ditrich, K.; Hoffmann, R. W. *Tetrahedron Lett.* **1985**, *26*, 6325. (b) Hanessian S.; Roy, P. J.; Petrini, M.; Hodgesi, P. J.; Di Fabrio, R.; Carganico, G. J. Org.
- Chem. 1990, 55, 5766.
  (65) Blanchette, M. A.; Choy, W.; Davis J. T.; Essenfeld, A. P.; Masamune, S.; Roush W. R.; Sakai, T. *Tetrahedron Lett.* 1984, 25, 2183.
- Luche, J.-L. *J. Am. Chem. Soc.* **1978**, *100*, 2226. Mitsunobu, O. *Synthesis* **1981**, 1. (66)
- (67)
- (68) The acylating agent was prepared from commercially available methyl ester by means of enzymatic hydrolysis with PLE (Fluka 46058) and subsequent reaction of the acid with oxalyl chloride. (a) Guindon, Y.; Therien, M.; Girard, Y.; Yoakim, Č.; Morton,
- (69)H. E. J. Org. Chem. **1984**, 49, 3912. (b) Guindon, Y.; Yoakim, C J. Org. Chem. **1987**, 52, 1680.
- (70) (a) Paterson, I.; Smith, J. D.; Ward, R. A. *Tetrahedron* 1995, 51, 9413. (b) Paterson, I.; Cumming, J. G.; Ward, R. A.; Lamboley, S. *Tetrahedron* 1995, 51, 9393. (c) Paterson, I.; Watson, C.; Yeung, K.-S.; Wallace, P. A.; Ward, R. A. J. Org. Cham 1997, 62, 452. Chem. 1997, 62, 452.
- (71) Luche, J.-L.; Petrier, C.; Lansard, J.-P.; Greene A. E. J. Org. *Chem.* 1983, *48*, 3837 and references therein.
  (72) Dossetter, A. G.; Jamison, T. F.; Jacobsen, E. N. *Angew. Chem.*,
- Int. Ed. 1999, 38, 2398.
- Saito, S.; Ishikawa, T.; Kuroda, A.; Koga, K.; Moriwake, T. (73)Tetrahedron 1992, 48, 4067.
- (74)(a) Paterson, I.; Goodman, J. M.; Lister, M. A.; Schumann, R. C.; McClure, C. K.; Norcross, R. D. Tetrahedron 1990, 46, 4663. (b) Paterson, I.; Oballa, R. M.; Norcross, R. D. Tetrahedron Lett. 1996, *37*, 8581
- (75)(a) Hibino, J.; Okazoe, T.; Takai, K.; Nozaki, H. Tetrahedron Lett. 1985, 26, 5579. (b) Takai, K.; Kakiuchi, T.; Kataoka, Y.; Utimoto, K. J. Org. Chem. 1994, 59, 2668.
- (76) (a) Bartlett, P. A.; Johnson, W. S.; Elliott, J. D. J. Am. Chem. Soc. 1983, 105, 2088. (b) Choi, V. M. F.; Elliott, J. D.; Johnson, W. S. Tetrahedron Lett. 1984, 25, 591. (c) Johnson, W. S.; Crackett, P. H.; Elliott, J. D.; Jagodzinski, J. J.; Lindell, S. D.; Natajaran, S. Tetrahedron Lett. 1984, 25, 3951.
- For a review on chiral acetals derived from optically active (77) alcohols, see: Alexakis, A.; Mangeney, P. Tetrahedron: Asymmetry 1990, 1, 477.
- For extensive studies on the mechanism and origin of stereose-(78) lective opening of chiral dioxane acetals and for leading references, see: (a) Denmark, S. E.; Willson, T. M.; Almstead, N. G. ences, see: (a) Denmark, S. E.; Willson, I. M.; Almstead, N. G.
   J. Am. Chem. Soc. **1989**, 111, 9258. (b) Denmark, S. E.;
   Almstead, N. G. J. Org. Chem. **1991**, 56, 6485. (c) Denmark, S.
   E.; Almstead, N. G. J. Am. Chem. Soc. **1991**, 113, 8089.
   (79) For recent examples, see: (a) Yamamoto, Y.; Abe, H.; Nishii, S.;
   Yamada, J.-i. J. Chem. Soc., Perkin Trans. 1 **1991**, 3253. (b)
   Chen, S.-H.; Sun, X.; Boyer, R.; Paschal, J.; Zeckner, D.; Current,
   W.; Zwaiéd, M.; Bodeingor, M. Bioarg Med. Chem. Lett. **2000**.
- W.; Zweifel, M.; Rodriguez, M. Bioorg. Med. Chem. Lett. 2000, 10. 2107.
- (80) (a) Scott, W.; McMurry, E. Acc. Chem. Res. 1988, 21, 47. (b) Comins, D. L.; Dchghani, A. Tetrahedron Lett. 1992, 33, 6299.
  (81) Stille, J. K.; Scott, J. W. J. Am. Chem. Soc. 1986, 108, 3033.
  (82) Collum, D. B.; McDonald, J. H.; Still, W. C. J. Am. Chem. Soc.
- 1980, 102, 2118.
- Rodriguez, A.; Nomen, M.; Spur, B. W.; Godfroid, J. J. Tetrahe-(83) dron Lett. 1999, 40, 5161 and references therein.
- (84)The olefination reagent was prepared in two steps by acylation of 2-TMS-ethanol with bromoacetyl bromide and treatment of the resulting bromoacetate with diphenyl phosphite in the presence of triethylamine.<sup>52</sup> The same Z/E-isomerization was painfully experienced also by
- (85)Ghosh, who subjected a close analogue of seco acid 168 to cyclization,  $^{\rm 17b,c}$  and by Paterson, during the synthesis of the  $C_{\rm 1} C_{20}$  core of **1**.<sup>15</sup>
- (86) Hosomi, A.; Sakurai, H. Tetrahedron Lett. 1976, 16, 1295
- (a) Mori, K.; Kuwahara, S. Tetrahedron 1982, 38, 521. (b) Mori, (87)K.; Kuwahara, S.; Ueda, H. Tetrahedron 1983, 39, 2439.
- (88) Schaus, S. E.; Branalt, J.; Jacobsen, E. N. J. Org. Chem. 1998, 63, 403.
- Behling, J. R.; Babiak, K. A.; Ng, J. S.; Campbell, A. L.; Moretti, R.; Koerner, M.; Lipshutz, B. H. *J. Am. Chem. Soc.* **1988**, *110*, (89)2641.
- Scott, J. W.; Crisp, G. T.; Stille, J. K. J. Am. Chem. Soc. 1984, (90)106, 4630.
- (91) (a) Narayanan, B. A.; Bunelle, W. H. Tetrahedron Lett. 1987, 28, 6261. (b) Lee, T. V.; Channon, J. A.; Cregg, C.; Porter, J. R.; Roden, F. S.; Yeoh, H. T. *Tetrahedron* **1989**, *45*, 5877. McDougal, P. G.; Rico, J. G.; Ph, Y.; Condon, B. D. *J. Org. Chem.*
- (92)1986, *51*, 3388.
- Molander, G. A.; Shakya, S. R. J. Org. Chem. 1996, 61, 5885.
  Hwu, J. R.; Jain, M. L.; Tsai, F.-Y.; Tsay, S.-C.; Balakumar, A.;
  Hakimelahi, G. H. J. Org. Chem. 2000, 65, 5077.
  Ishihara, K.; Mouri, M.; Gao, Q.; Maruyama, T.; Furuta, K.;
  Yamamoto, H. J. Am. Chem. Soc. 1993, 115, 11490. (93)(94)
- (95)

- (96) Taber, D. F.; Song, Y. J. Org. Chem. 1996, 61, 7508.
- (97) Müller, S.; Liepold, B.; Roth, G. J.; Bestmann, H. J. Synlett 1996, 521
- (98) Crimmins, M. T.; Emmitte, K. A.; Katz, J. D. Org. Lett. 2000, 2, 2165.
- (99) For previous reports on citronellal-based syntheses of laulimalide (100) Prashad, M.; Har, D.; Kim, H.-Y.; Repic, O. *Tetrahedron Lett.*
- 1998. 39. 7067.
- (101) The same reaction sequence using the phenylalanine-derived Evans auxiliary was previously reported by a Korean group (cf. Scheme 52).16b
- (102) Lida, H.; Yamazaki, N.; Kibayashi, C. J. Org. Chem. 1986, 51, 1069. For an analogous anti-selective reduction with zinc borohydride, see Scheme 22.
- (103) Crimmins, M. T.; King, B. W. J. Am. Chem. Soc. 1998, 120, 9084. (104) For a review, see: Yamamoto, Y. Angew. Chem., Int. Ed. Engl.
- 1986, 25, 947. (105) For related studies, see: Marshall, J. A. Chem. Rev. 1996, 96,
- 31. (a) Wipf, P.; Xu, W. Tetrahedron Lett. 1994, 35, 5197. (b) Wipf, (106)
- P.; Ribe S. J. Org. Chem. 1998, 63, 6454. (107) Masamune, S.; Ma, P.; Okumoto, H.; Ellingboe, J. W.; Ito, Y. J.
- Org. Chem. 1984, 49, 2834. (108) (a) Nelson, S. G.; Peelen, T. J.; Wan, Z. J. Am. Chem. Soc. 1999,
- 121, 9742. (b) Nelson, S. G.; Wan, Z. Org. Lett. 2000, 2, 1883. (109) Zipp, G. G.; Hilfiker, M. A.; Nelson, S. G. Org. Lett. 2002, 4, 1823.
- (110)Shimizu, T.; Osako, K.; Nakata, T. Tetrahedron Lett. 1997, 38, 2685.
- (111) Nelson, S. G.; Wan, Z.; Stan, M. A. J. Org. Chem. 2002, 67, 4680.
- (112) A similar C-propargylation was performed by Williams (cf. Scheme 38).
- (113) Schrock, R. R.; Murdzek, J. S.; Bazan, G. C.; Robbins, J.; DiMare, M.; O'Regan, M. J. Am. Chem. Soc. 1990, 112, 3875.

- (114) Corey, E. J.; Lee, D.-H. Tetrahedron Lett. 1993, 34, 1737.
- (115) For additional involvement of 213 in the synthesis of laulimalide fragments, see Schemes 34, 48, and 49.
- (116)Keck, G. E.; Li, X. Y.; Krishnamurthy, D. J. Org. Chem. 1995, 60, 5998.
- (117) (a) Keck, G. E.; Tarbet, K. H.; Geraci, L. S *Tetrahedron Lett.* 1993, 34, 7827. (b) Keck, G. E.; Tarbet, K. H.; Geraci, L. S. J. Am. Chem. Soc. 1993, 115, 8467.
  (118) Trost, B. M.; Herndon, J. W. J. Am. Chem. Soc. 1984, 106, 6835.
- (119) Williams, D. R.; Kissel, W. S. J. Am. Chem. Soc. 1998, 120,
- 11198.
- (120) For similar approaches to generate the stereochemistry at C<sub>20</sub>, see: Schemes 40 and 43.
- (121) Takai, K.; Nitta, K.; Utimoto, K. J. Am. Chem. Soc. 1986, 108, 7408
- (122)Le Merrer, Y.; Gravier-Pelletier, C.; Dumas, J.; Depezay, J.-C. Tetrahedron Lett. 1990, 31, 1003.
- (123) Borredon, E.; Delmas, M.; Gaset, A. Tetrahedron Lett. 1982, 23, 5283.
- (124) (a) Jacobsen, E. N.; Tokunaga, M.; Larrow, J. F.; Kakiuchi, F. Science 1997, 277, 936. (b) Jacobsen, E. N.; Furrow, M. E.; Schaus, S. E. J. Org. Chem. 1998, 63, 6776. (c) Jacobsen, E. N.;
- Schaus, S. E. J. Org. Chem. 1998, 63, 2164.
  (125) For related protocols, see: (a) Takano, S.; Shimazaki, Y.; Iwabuchi, Y.; Ogasawara, K. Tetrahedron Lett. 1990, 31, 3619.
  (b) Oizumi, M.; Takahashi, M.; Ogasawara, K. Synlett 1997, 1111.
- (126) Burke, S. D.; Ng, R. A.; Morrison, J. A.; Alberti, M. J. J. Org. Chem. 1998, 63, 3160.
- Evans, D. A.; Morrissey, M. M.; Dorrow, R. L. J. Am. Chem. Soc. (127)1985, 107, 4346.
- (128)For an analogous approach in Crimmins' total synthesis, see: Scheme 35.

CR940368C